

# ESE370: Circuit-Level Modeling, Design, and Optimization for Digital Systems

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Lec 6: September 17, 2021  
MOS Model



# You are Here: Transistor Edition

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- Previously: simple models (0<sup>th</sup> and 1<sup>st</sup> order)
  - Comfortable with basic functions and circuits
- This lecture and next week(4 lectures)
  - Detailed semiconductor discussion
  - MOSFET phenomenology
  - Don't Blink!
- Rest of term
  - Implications of the MOS device



# Today

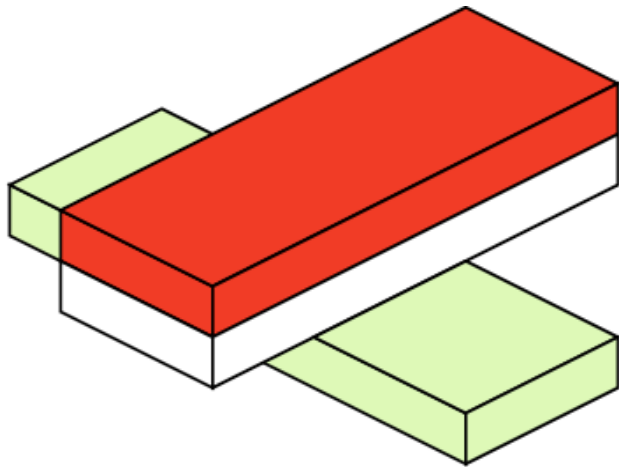
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- ❑ MOS Structure
- ❑ Basic Fabrication
- ❑ Semiconductor Physics
  - Metals, insulators
  - Silicon lattice
  - Band gaps
  - Doping
  - Field Effects

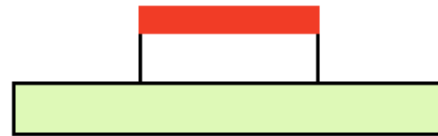


# MOS

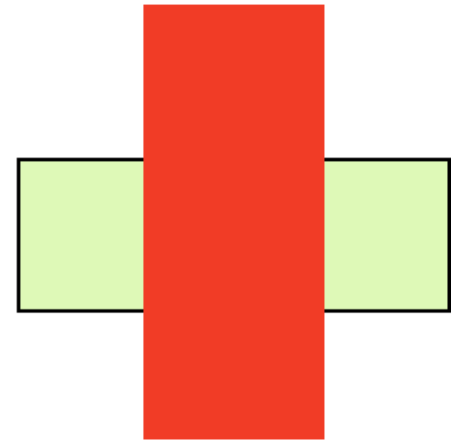
□ Metal Oxide Semiconductor



**Oblique**



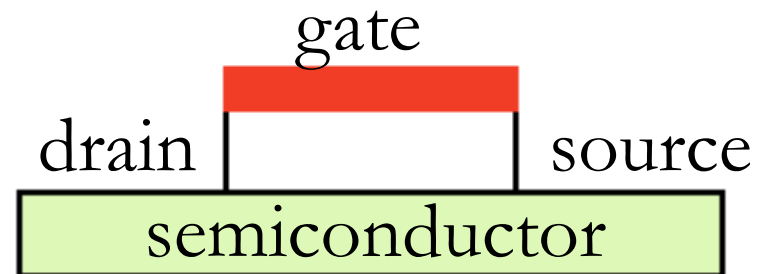
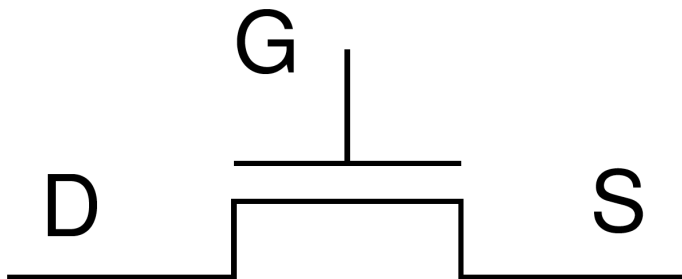
**Side**



**Top**

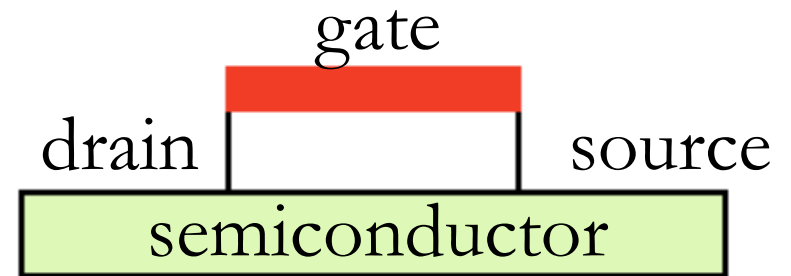
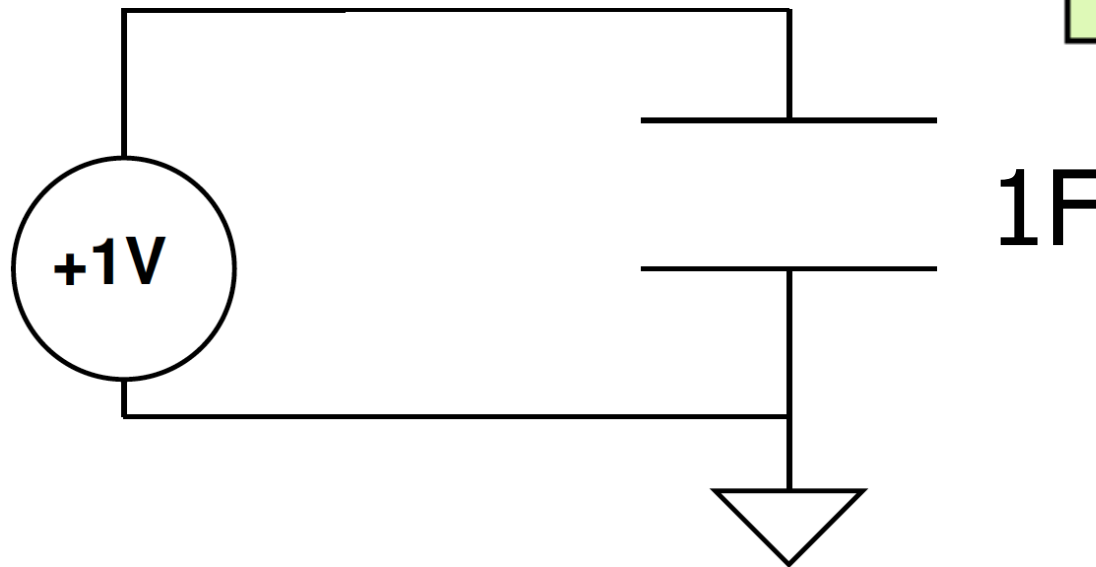
# MOS

- ❑ **M**etal – gate
- ❑ **O**xide – insulator separating gate from semiconductor
  - Ideally: no conduction from gate to semiconductor
- ❑ **S**emiconductor – between source and drain
- ❑ See why gate input is capacitive?



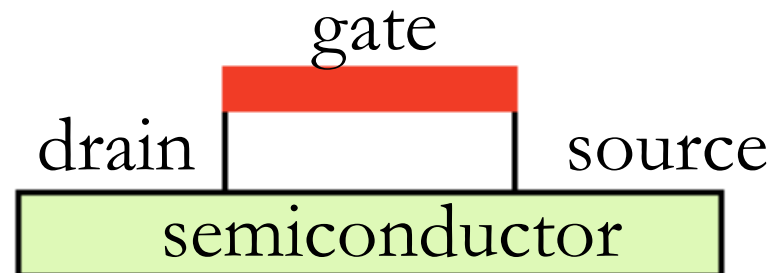
# (MOS) Capacitor (preclass 1)

- Charge distribution and field?
- How much charge on plates?



# Idea

- ❑ Semiconductor – can behave as metal or insulator
- ❑ Voltage on gate induces an electrical field
- ❑ Induced field attracts (repels) charge in semiconductor to form a channel
  - Semiconductor can be switched between conducting and not conducting
  - Hence “Field-Effect” Transistor

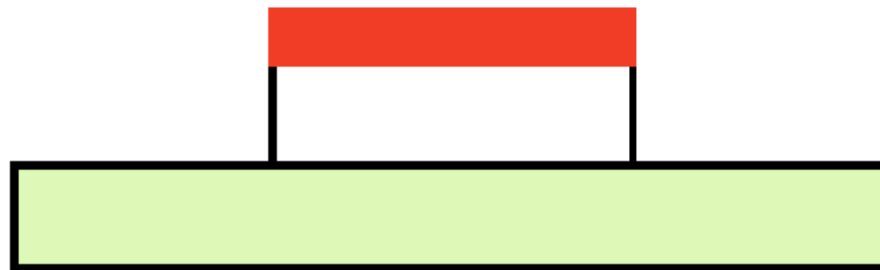


# Source/Drain Contacts

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- Contacts: Conductors → metallic
  - Connect to metal wires that connect transistors



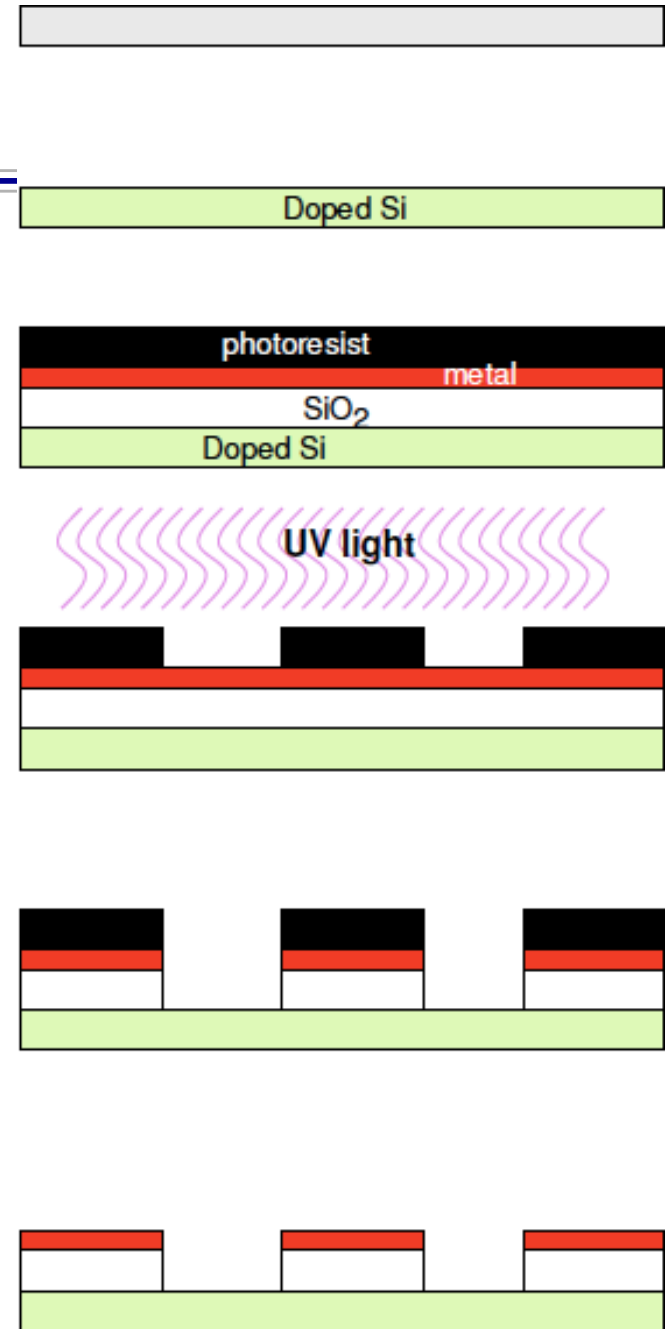


# Fabrication

- ❑ Start with Silicon wafer
- ❑ Dope silicon
- ❑ Grow Oxide ( $\text{SiO}_2$ )
- ❑ Deposit Metal
- ❑ Photoresist mask and etch to define where features go

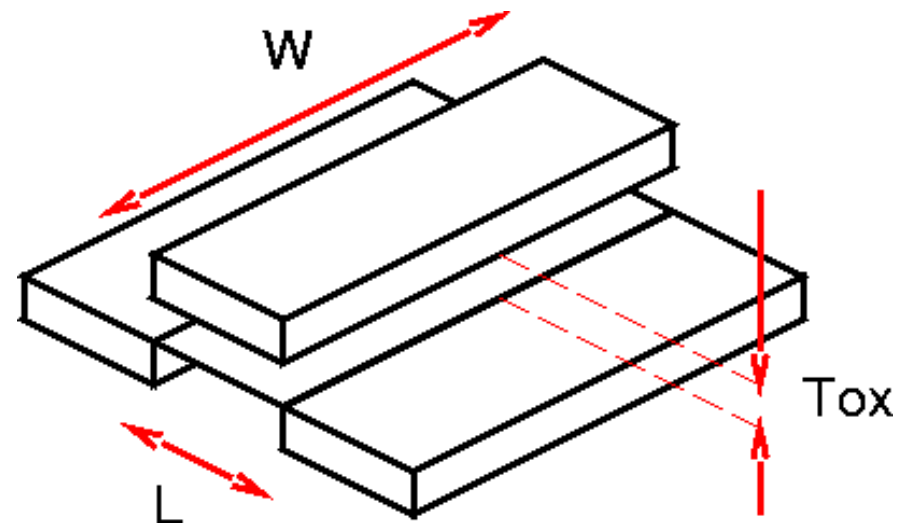
<https://youtu.be/35jWSQXku74?t=119>

Time Code: 2:00-4:30



# Dimensions

- ❑ Channel Length ( $L$ )
- ❑ Channel Width ( $W$ )
- ❑ Oxide Thickness ( $T_{ox}$ )
  
- ❑ Process named by minimum length
  - $22\text{nm} \rightarrow L=22\text{nm}$



# Semiconductor Physics

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# Conduction

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- ❑ Metal – conducts
- ❑ Insulator – does not conduct
- ❑ Semiconductor – can act as either

# Why does metal conduct? (preclass 2)

Group →	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
↓ Period																		
1	1 H																	2 He
2	3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
3	11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
4	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
6	55 Cs	56 Ba		72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
7	87 Fr	88 Ra		104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Uub	113 Uut	114 Uuq	115 Uup	116 Uuh	117 Uus	118 Uuo
Lanthanides				57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
Actinides				89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

<http://chemistry.about.com/od/imagesclipartstructures/ig/Science-Pictures/Periodic-Table-of-the-Elements.htm>

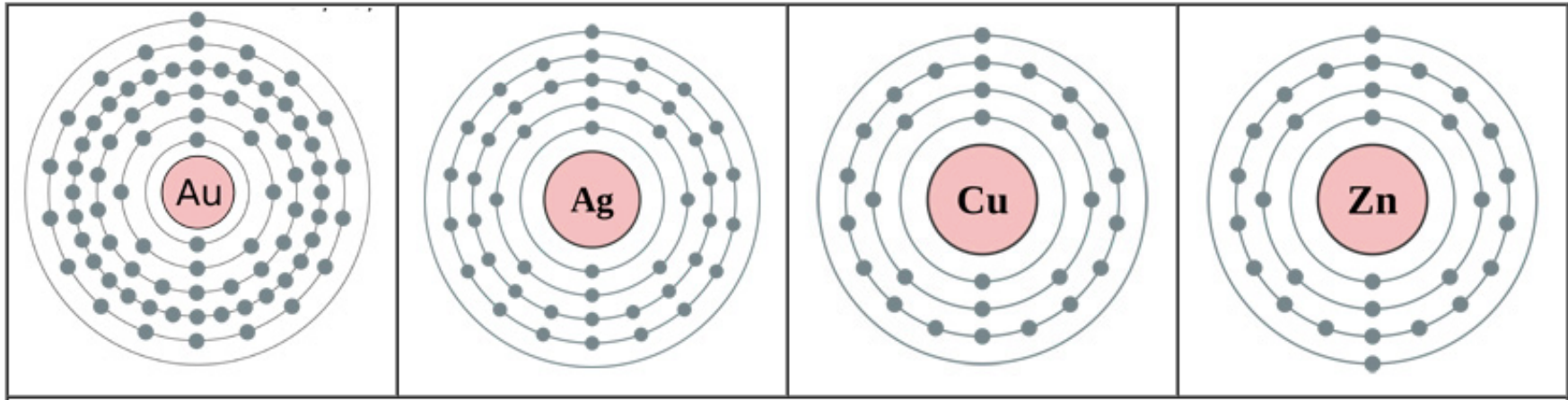
# Why does metal conduct? (preclass 2)

Gold

Silver

Copper

Zinc

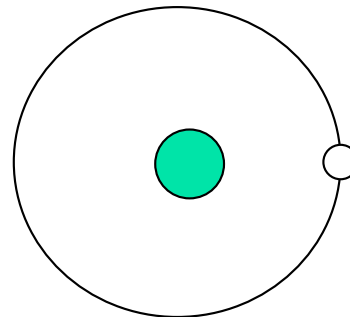




# Conduction

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- ❑ Electrons move
- ❑ Must be able to “remove” electron from atom or molecule

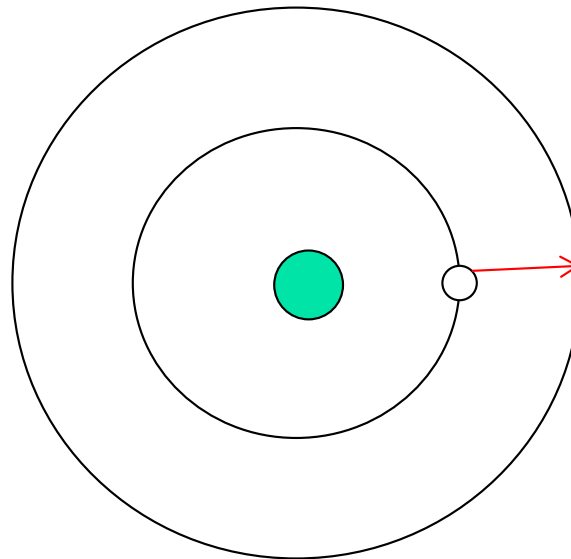




# Atomic States

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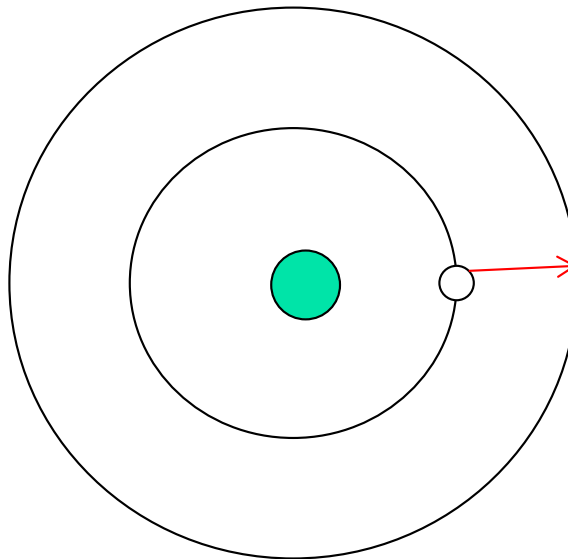
- ❑ Quantized Energy Levels (bands)
  - Valence and Conduction Bands
- ❑ Must have enough energy to change level (state)





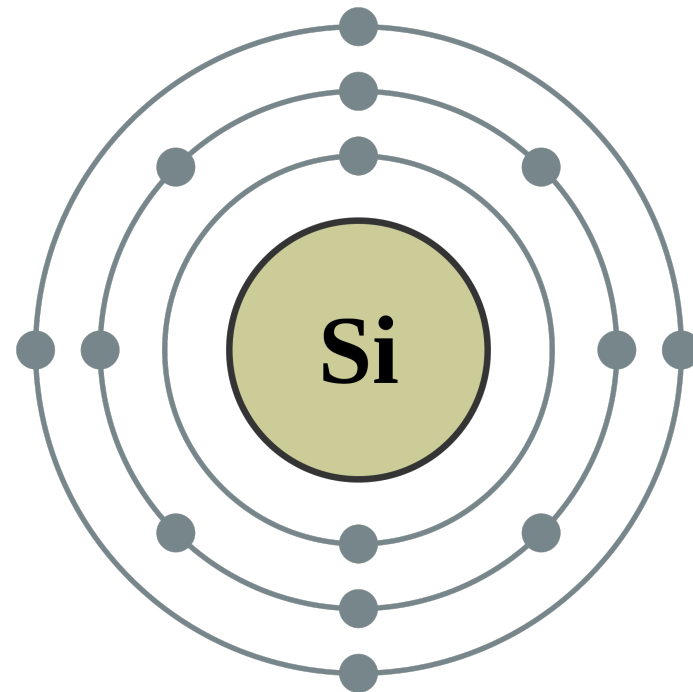
# Thermal Energy

- Except at absolute 0
  - There is always free energy
  - Causes electrons to hop around
    - ....when there is enough energy to change states
  - Energy gap between states determines energy required



# Silicon Atom (preclass 3)

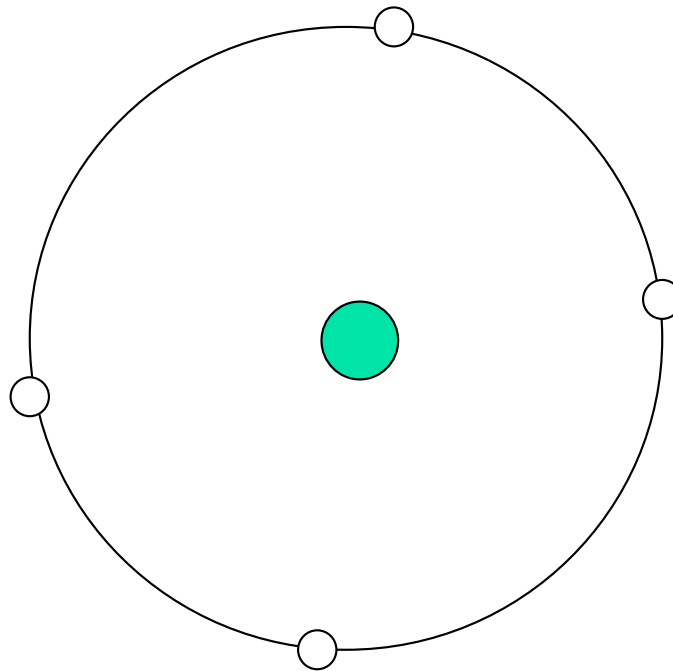
- How many valence electrons?



# Silicon

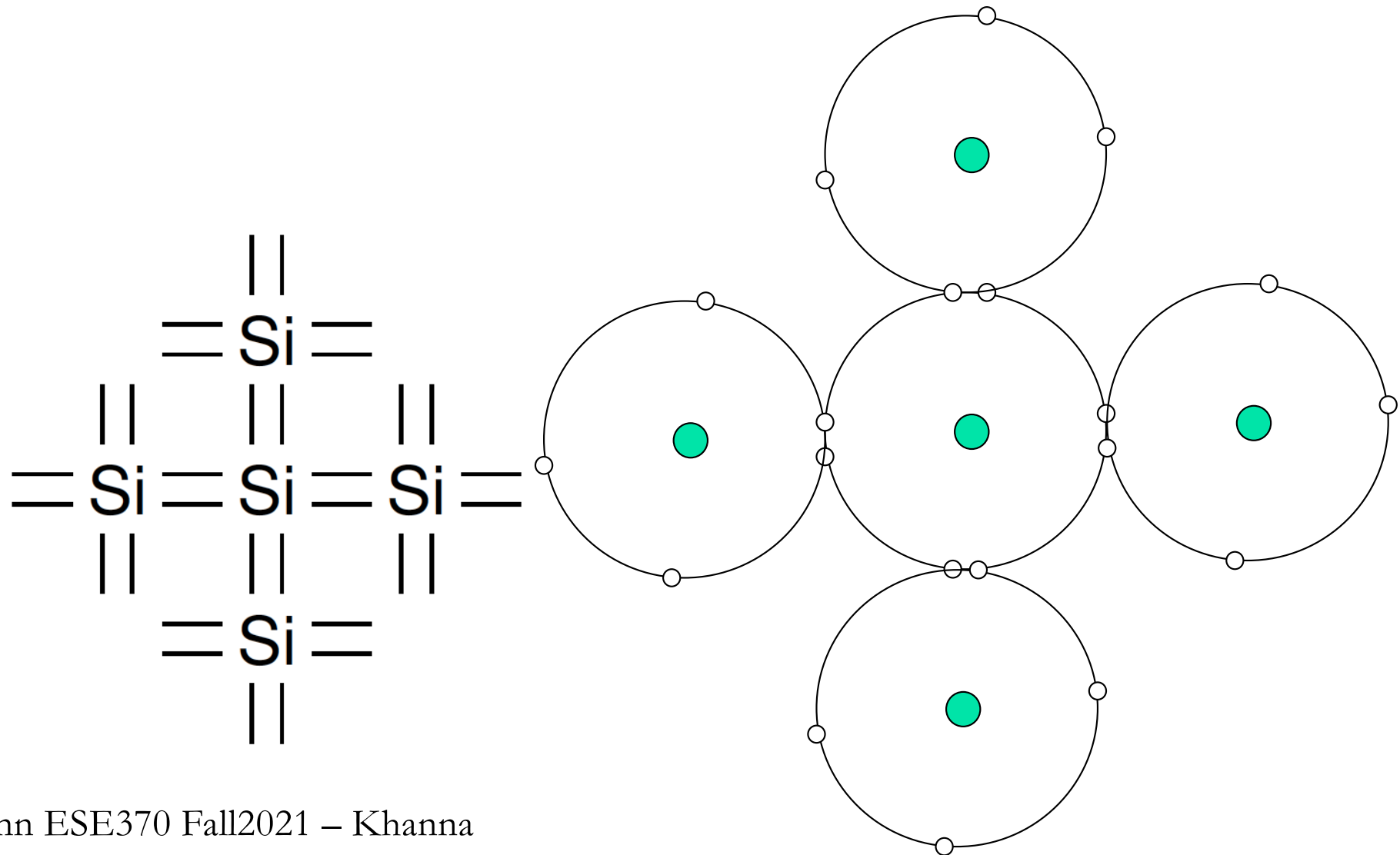
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- 4 valence electrons
  - Inner shells filled
  - Only outer shells contribute to chemical interactions



# Silicon-Silicon Bonding

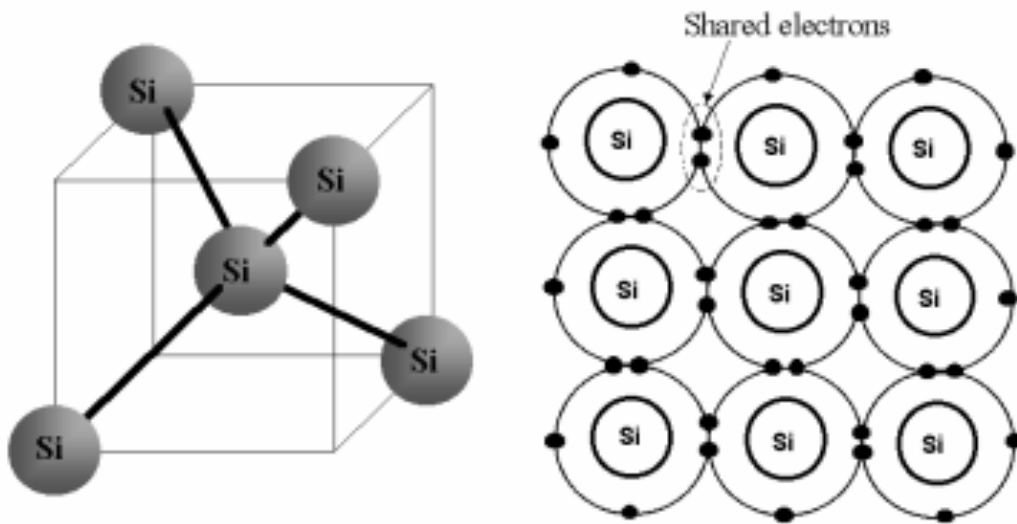
- Can form covalent bonds with 4 other silicon atoms



# Silicon Lattice

- Forms into crystal lattice

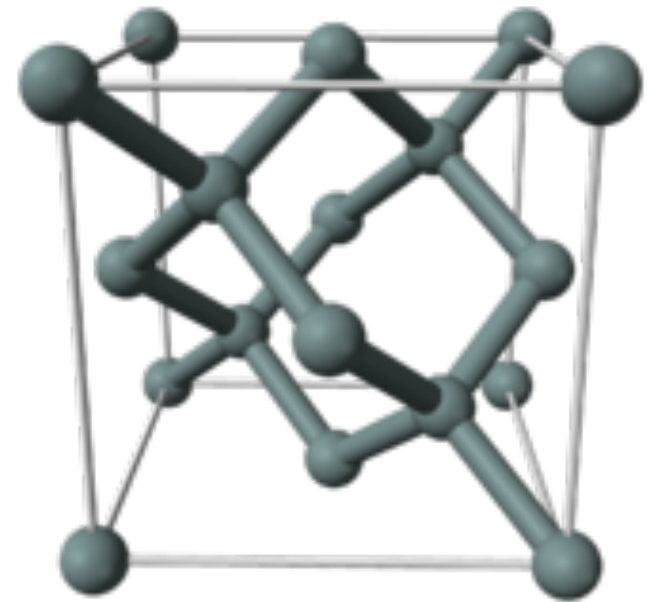
## Crystal Structure of Single Crystal Silicon



Hong Xiao, Ph. D.

[www2.austin.cc.tx.us/HongXiao/Book.htm](http://www2.austin.cc.tx.us/HongXiao/Book.htm)

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# Silicon Ingot

1 impurity atom  
per 10 billion  
silicon atoms

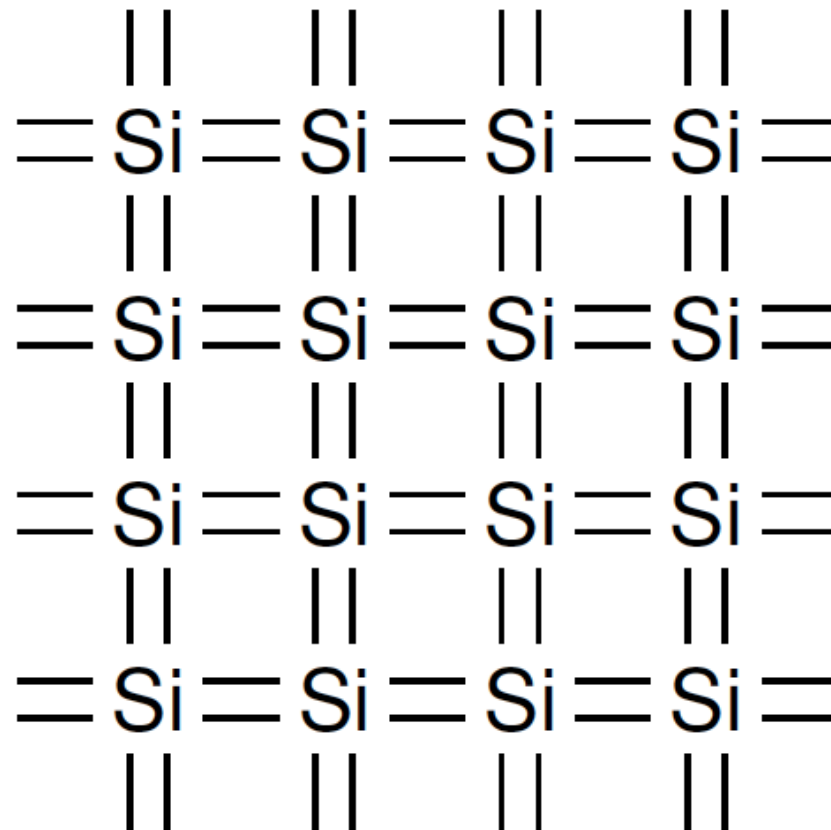




# Silicon Lattice

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- Cartoon two-dimensional view

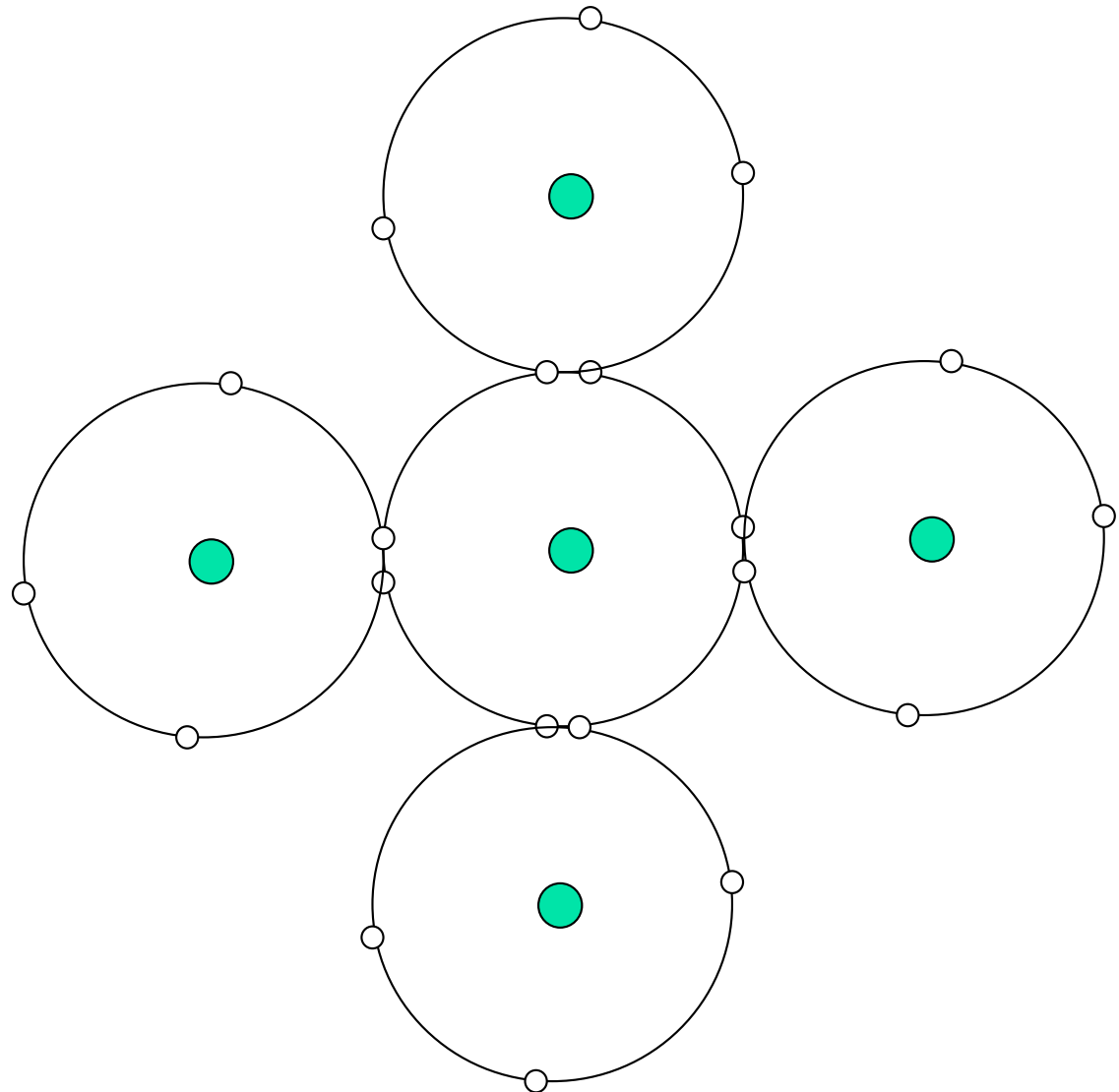




# Outer Orbital?

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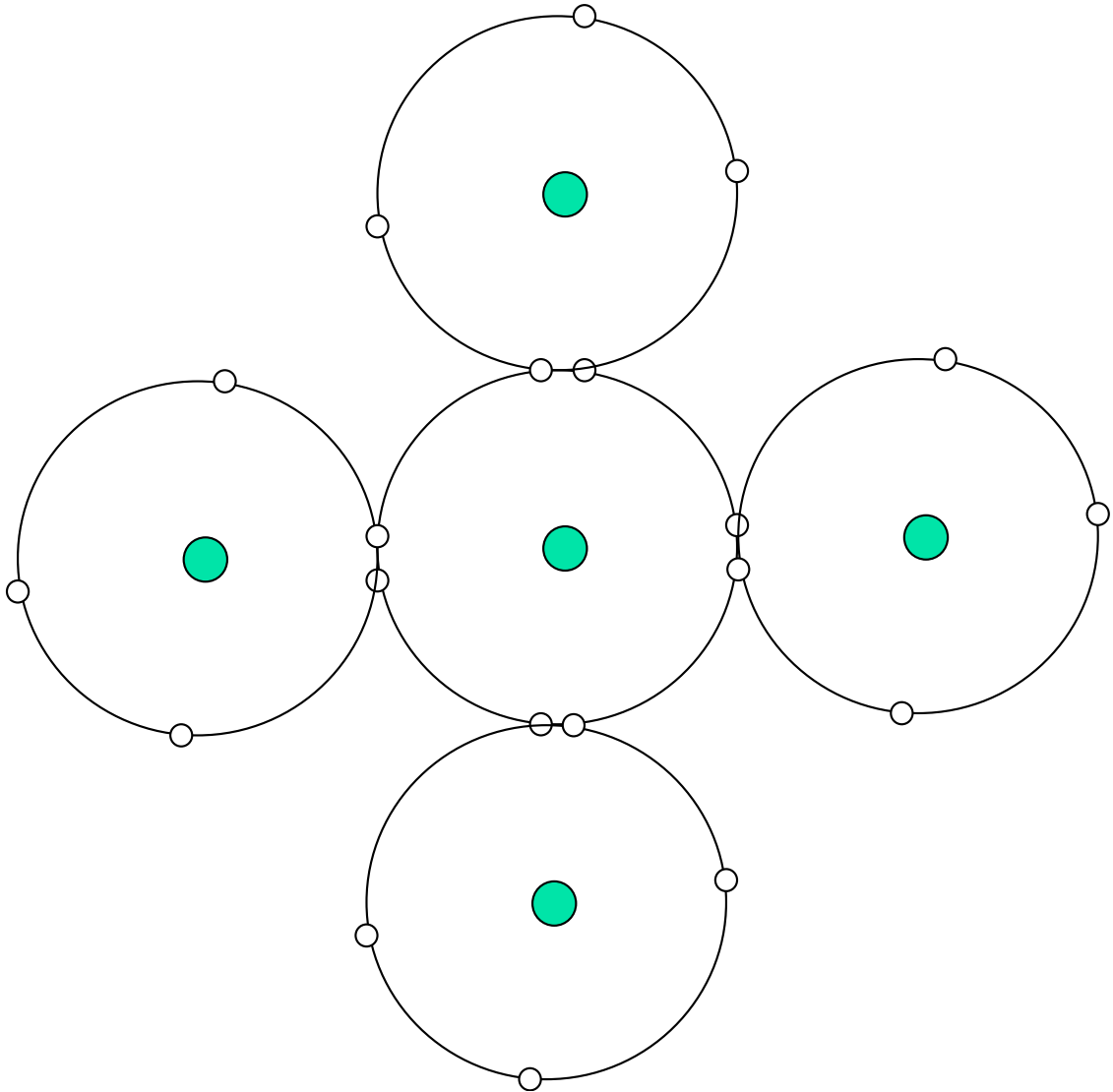
- What happens to outer shell in Silicon lattice?





# Energy?

- What does this say about energy to move electron?





# Energy State View

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Energy

Valance Band – all states filled



# Energy State View

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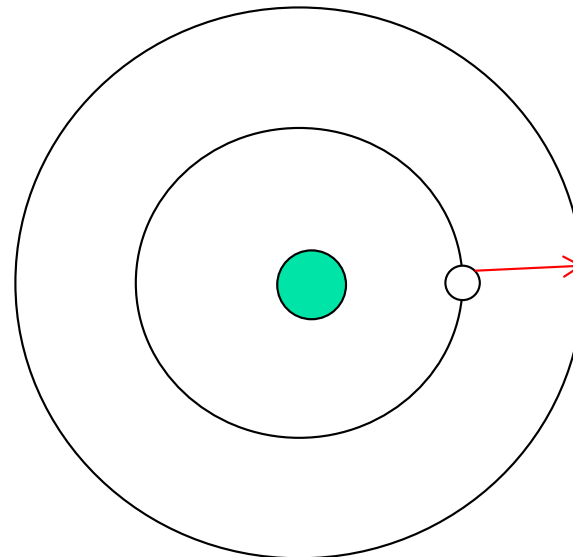
Conduction Band– all states empty

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Energy

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Valance Band – all states filled





# Energy State View

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Conduction Band— all states empty

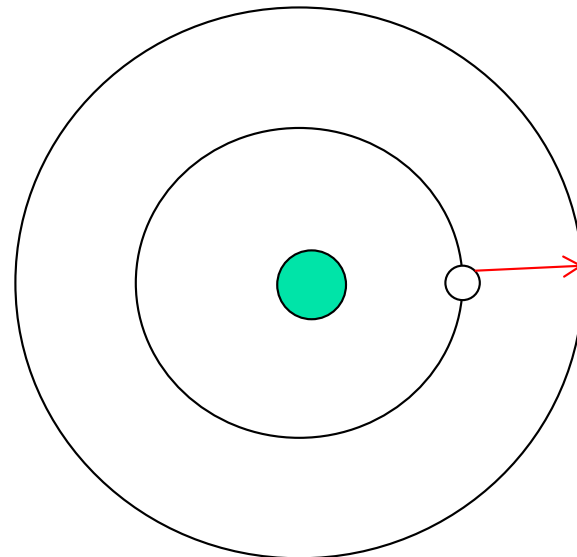
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Energy

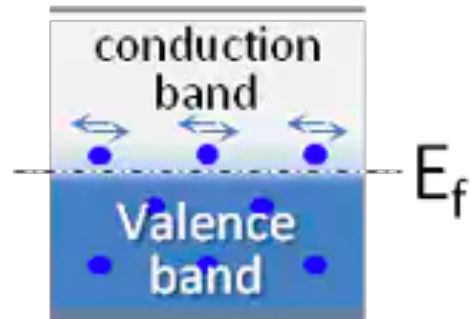
Band Gap

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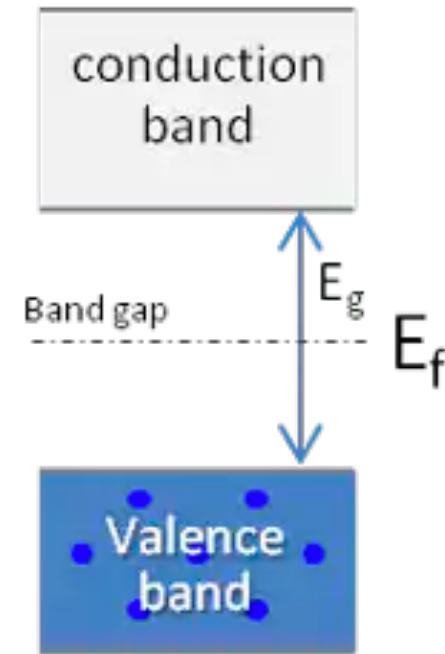
Valance Band – all states filled



# Band Gap and Conduction

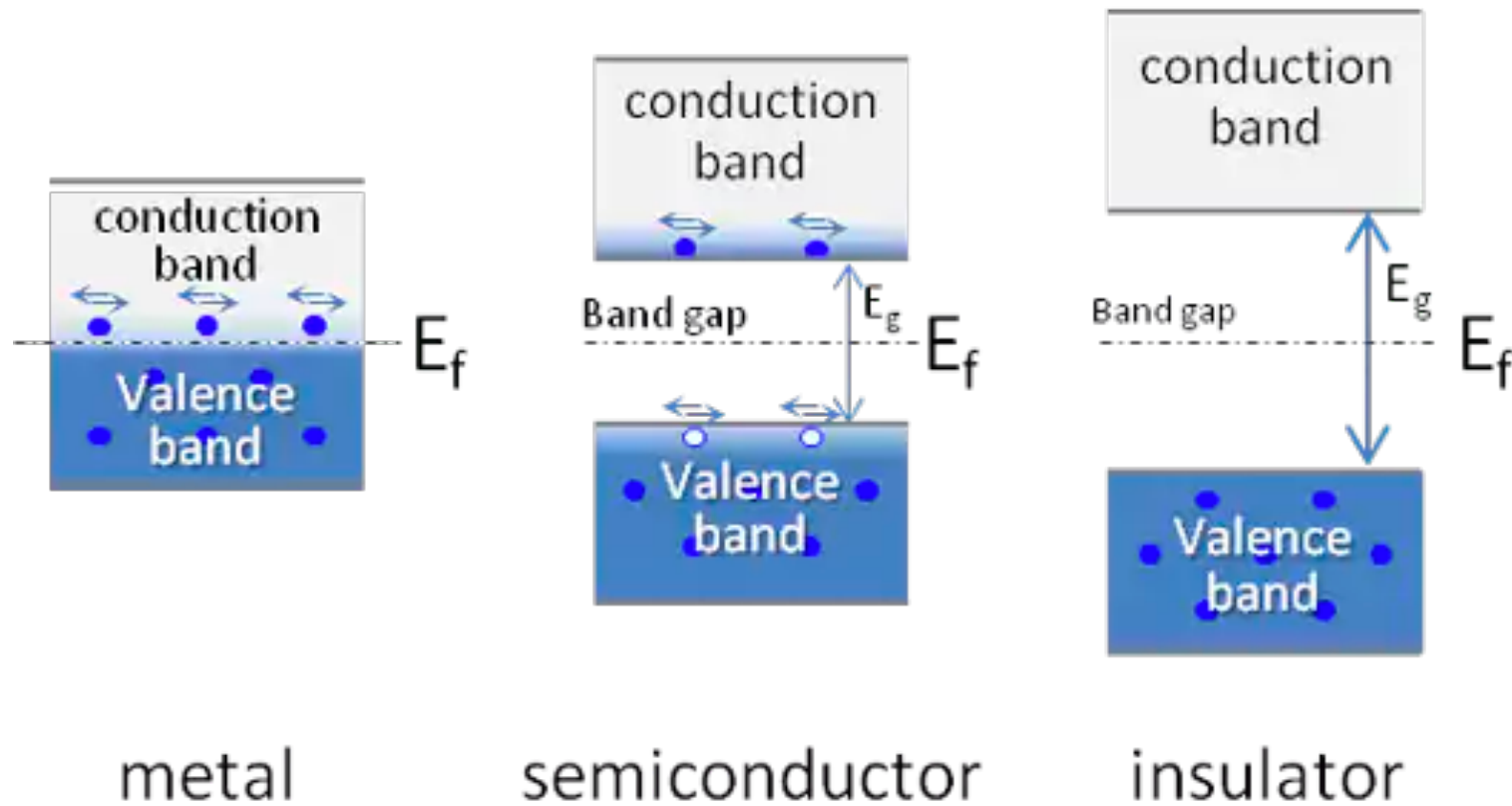


metal



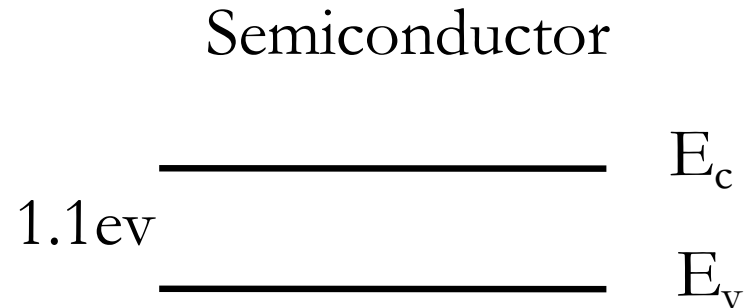
insulator

# Band Gap and Conduction



# Band Gap and Conduction

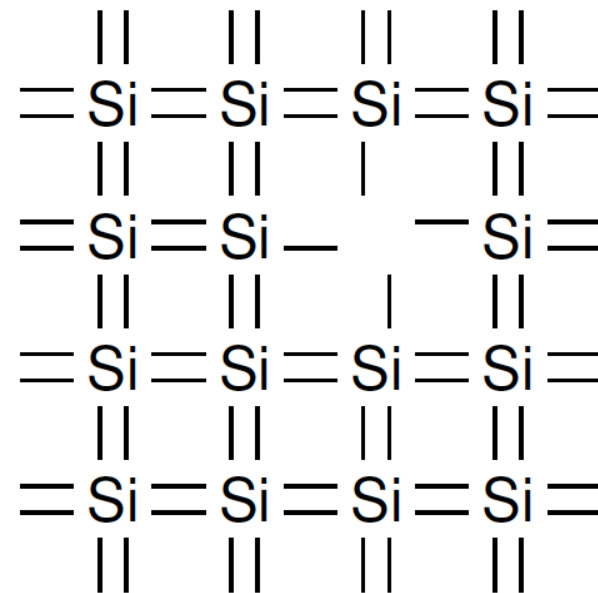
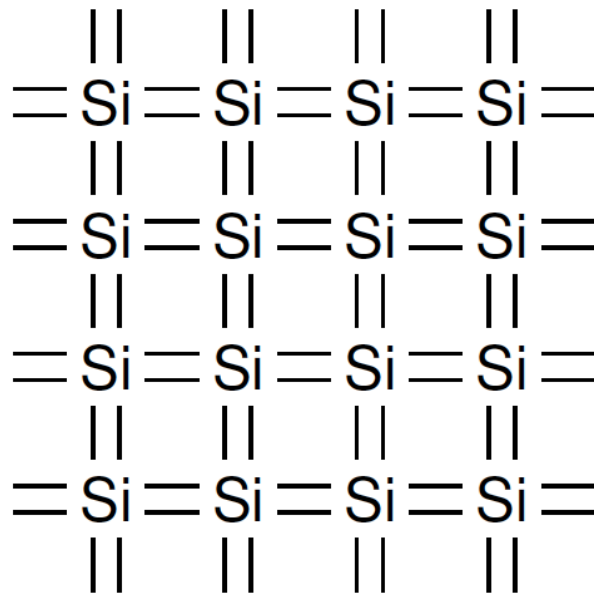
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$$1\text{eV} = 160 \text{ zeptojoules } (10^{-21} \text{ J})$$

# Doping

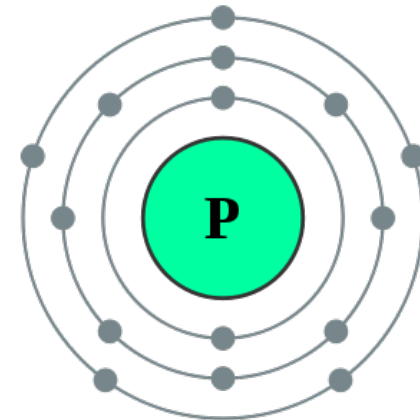
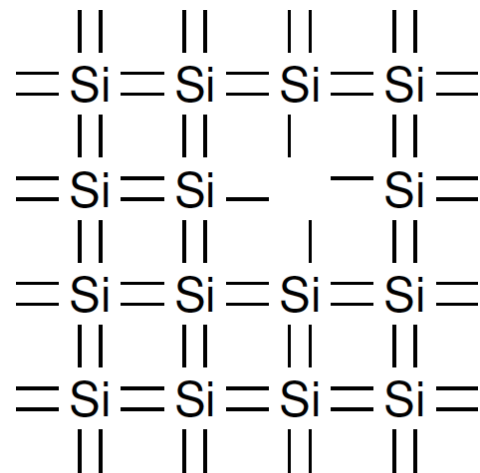
- Add impurities to Silicon Lattice
  - Replace a Si atom at a lattice site with another





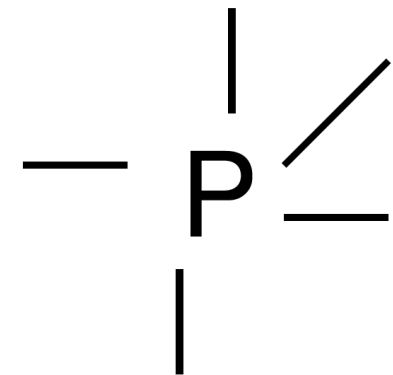
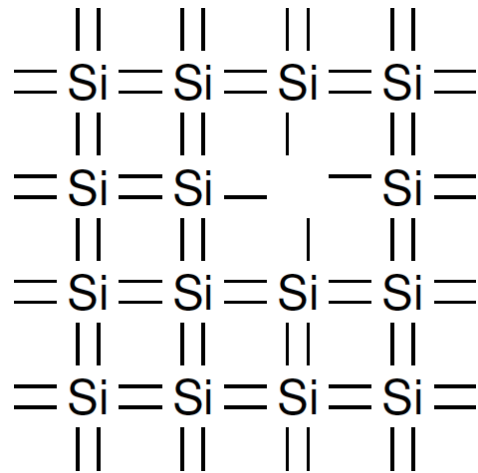
# Doping

- Add impurities to Silicon Lattice
  - Replace a Si atom at a lattice site with another
- Add a Group 15 element
  - *E.g.* P (Phosphorus)
  - How many valence electrons?



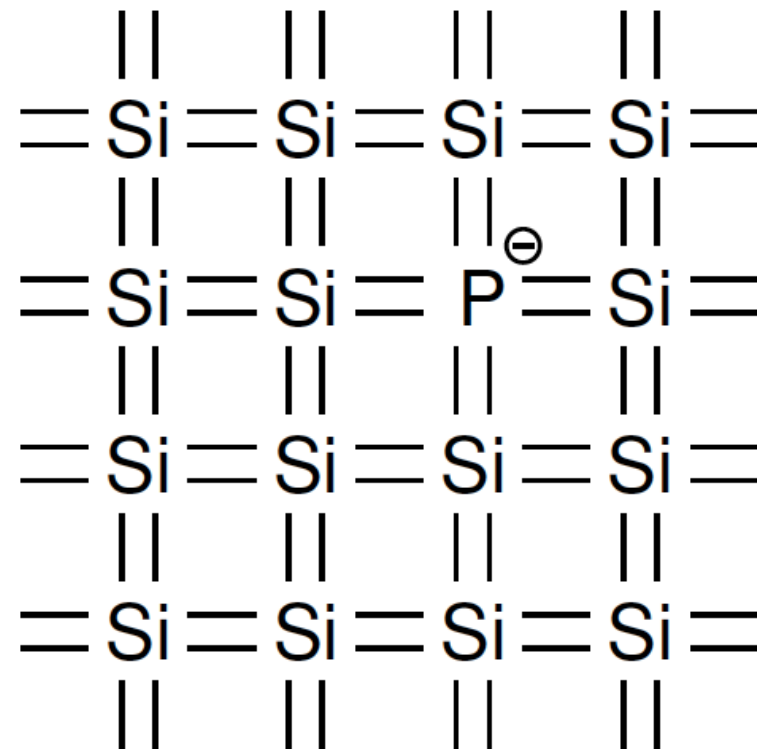
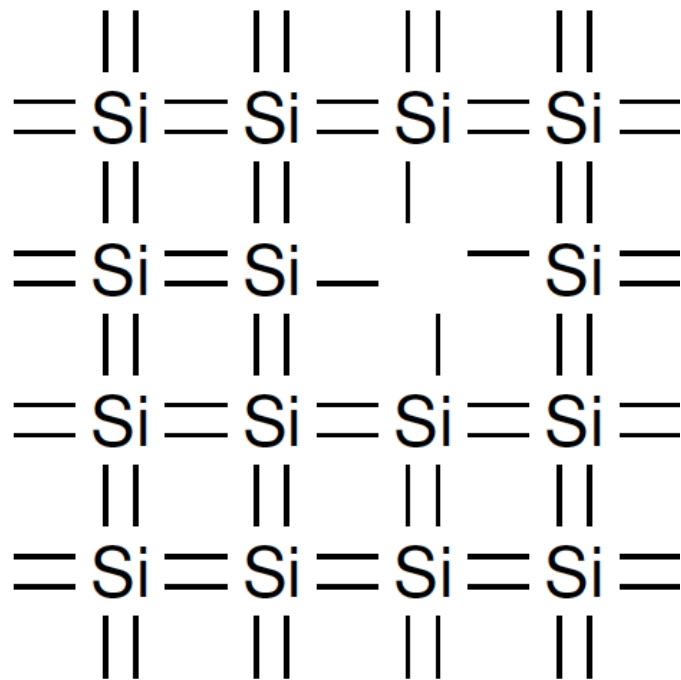
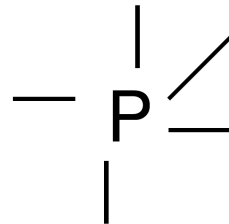
# Doping

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# Doping with P

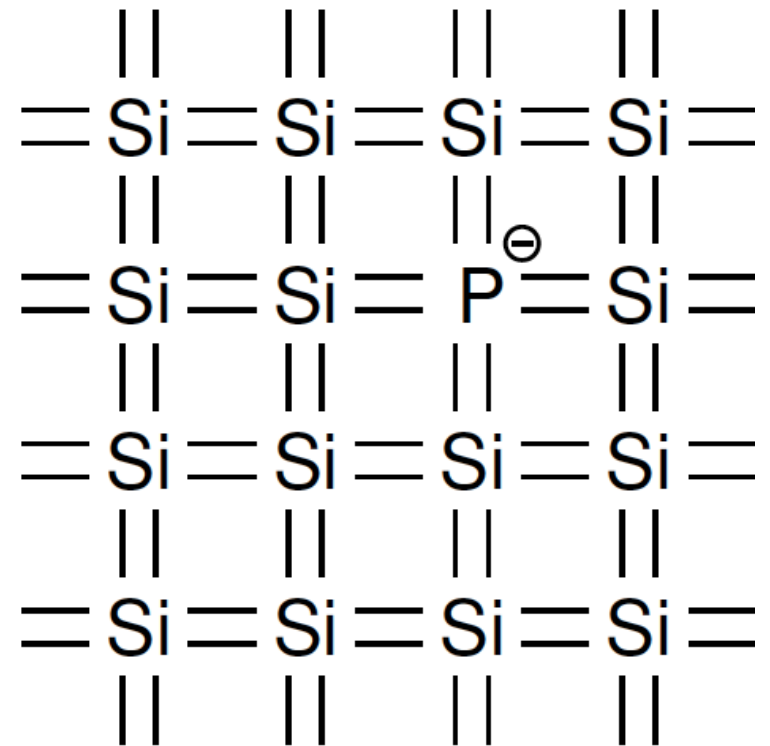




# Doping with P

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- ❑ End up with extra electrons
  - Donor electrons
- ❑ Not tightly bound to atom
  - Low energy to displace
  - Easy for these electrons to move

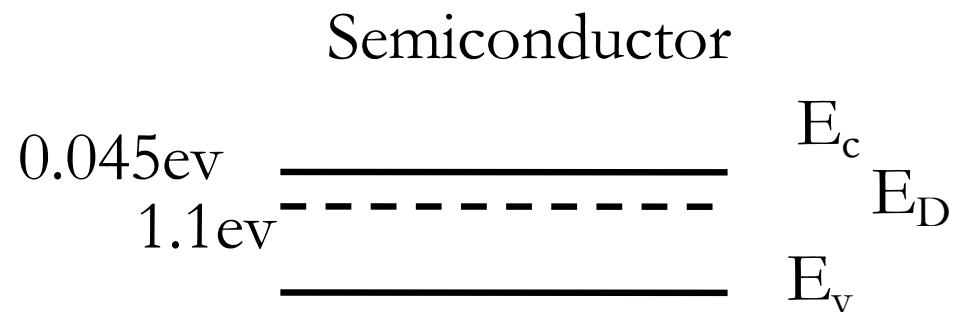




# Doped Band Gaps

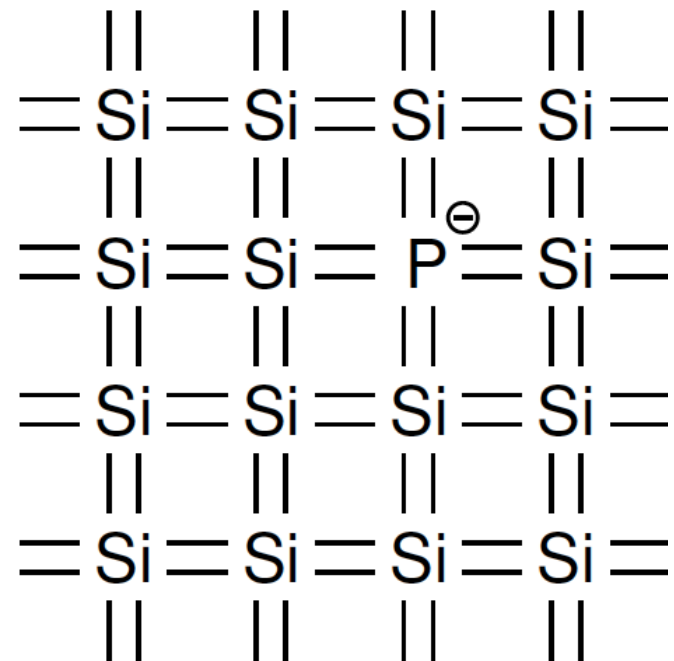
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- Addition of donor electrons makes more metallic
  - Easier to conduct



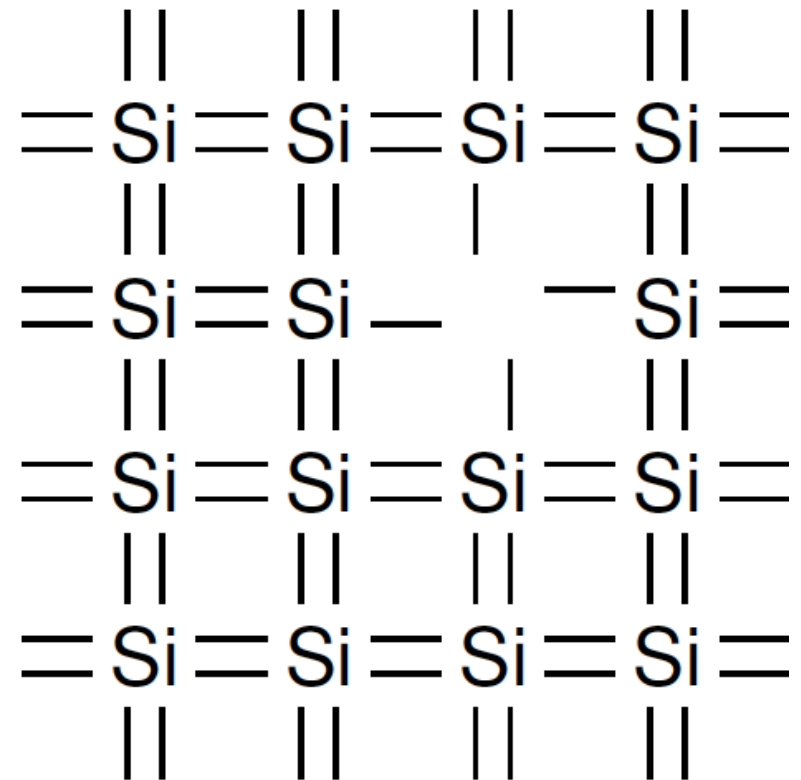
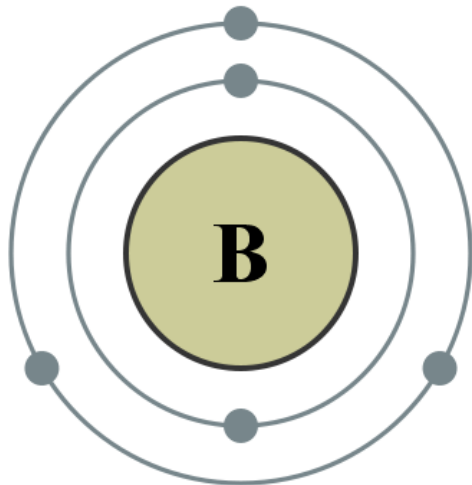
# Localized

- ❑ Donor electron is localized
  - Won't go far if no low energy states nearby
- ❑ Increasing doping concentration
  - Ratio of P atoms to Si atoms
  - Decreases energy to conduct



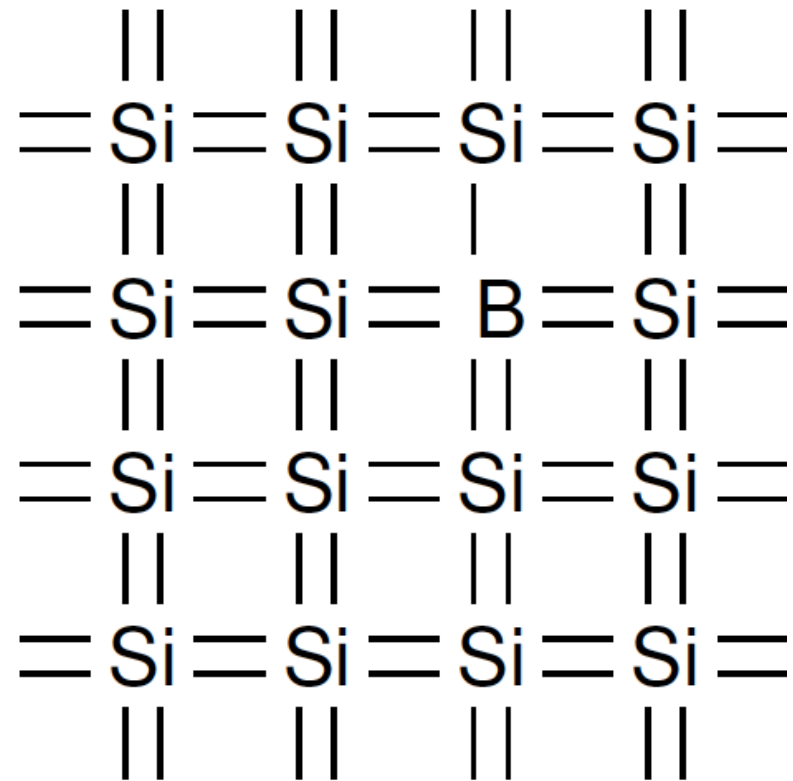
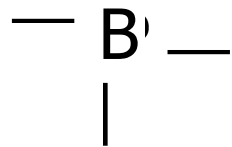
# Doping

- What happens if we replace Si atoms with group 13 atom instead?
  - E.g. B (Boron)
  - Valance band electrons?



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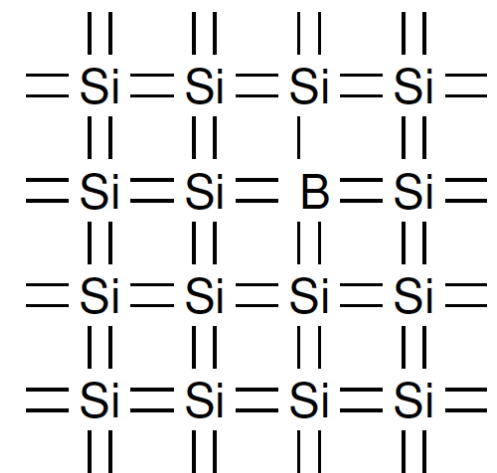




# Doping with B

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- End up with electron vacancies -- Holes
  - Acceptor electron sites
- Easy for electrons to shift into these sites
  - Low energy to displace
  - Easy for the electrons to move
    - Movement of an electron best viewed as movement of hole

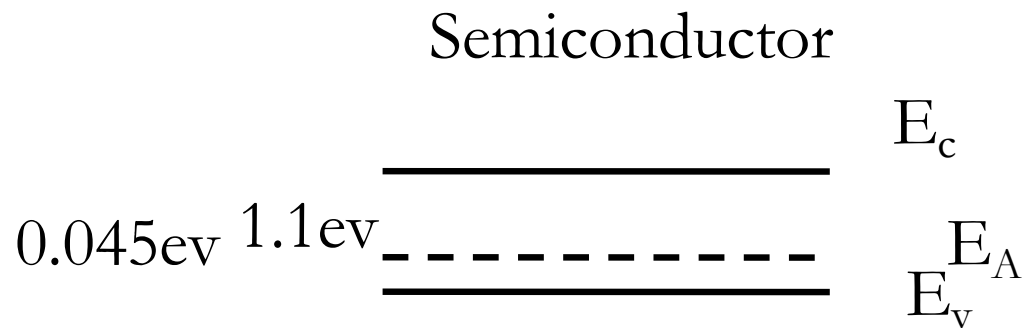




# Doped Band Gaps

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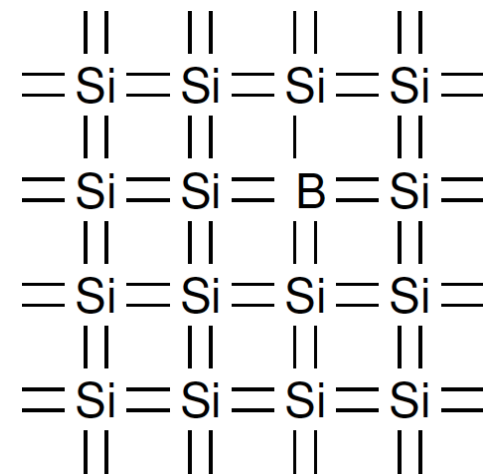
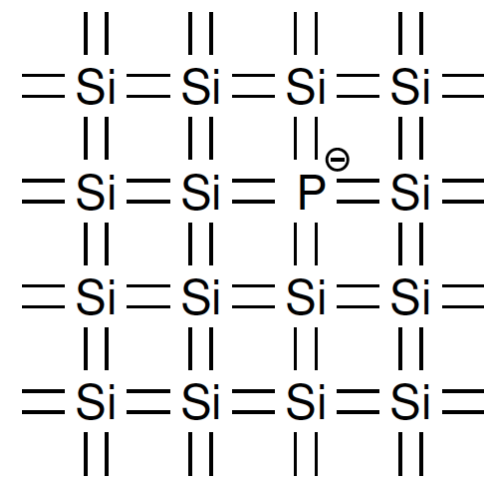
- Addition of acceptor sites makes more metallic
  - Easier to conduct





# MOSFETs

- Donor doping
  - Excess electrons
  - Negative or N-type material
  - NFET
  
- Acceptor doping
  - Excess holes
  - Positive or P-type material
  - PFET

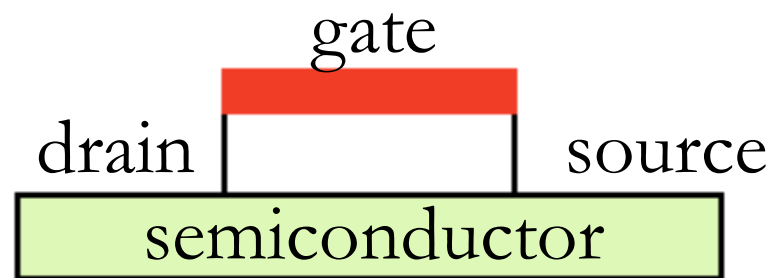
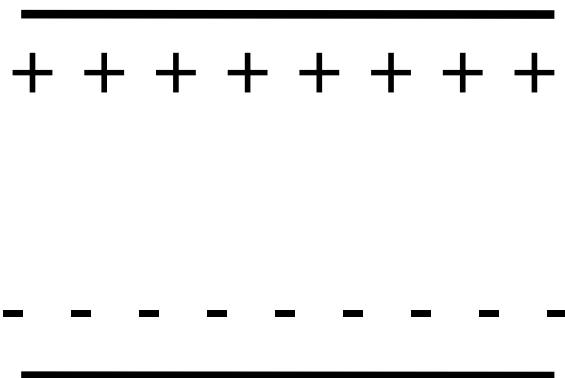




# Capacitor Charge

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- Remember capacitor charge

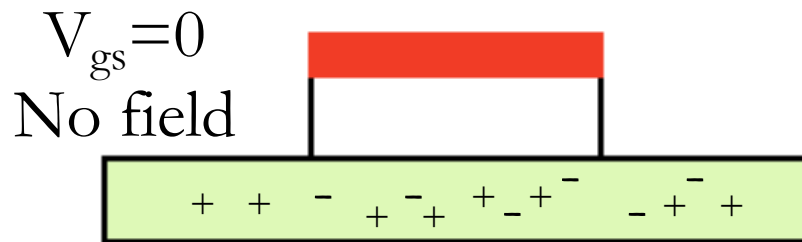




# MOS Field?

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- What does “capacitor” field do to the doped semiconductor channel?

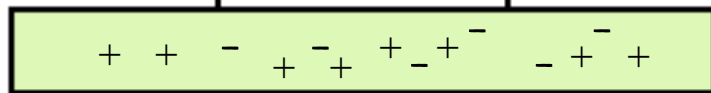




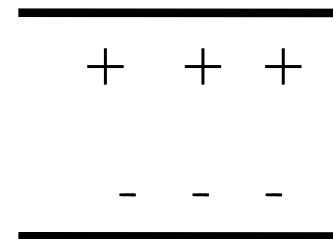
# MOS Field?

- What does “capacitor” field do to the doped semiconductor channel?

$V_{gs} = 0$   
No field



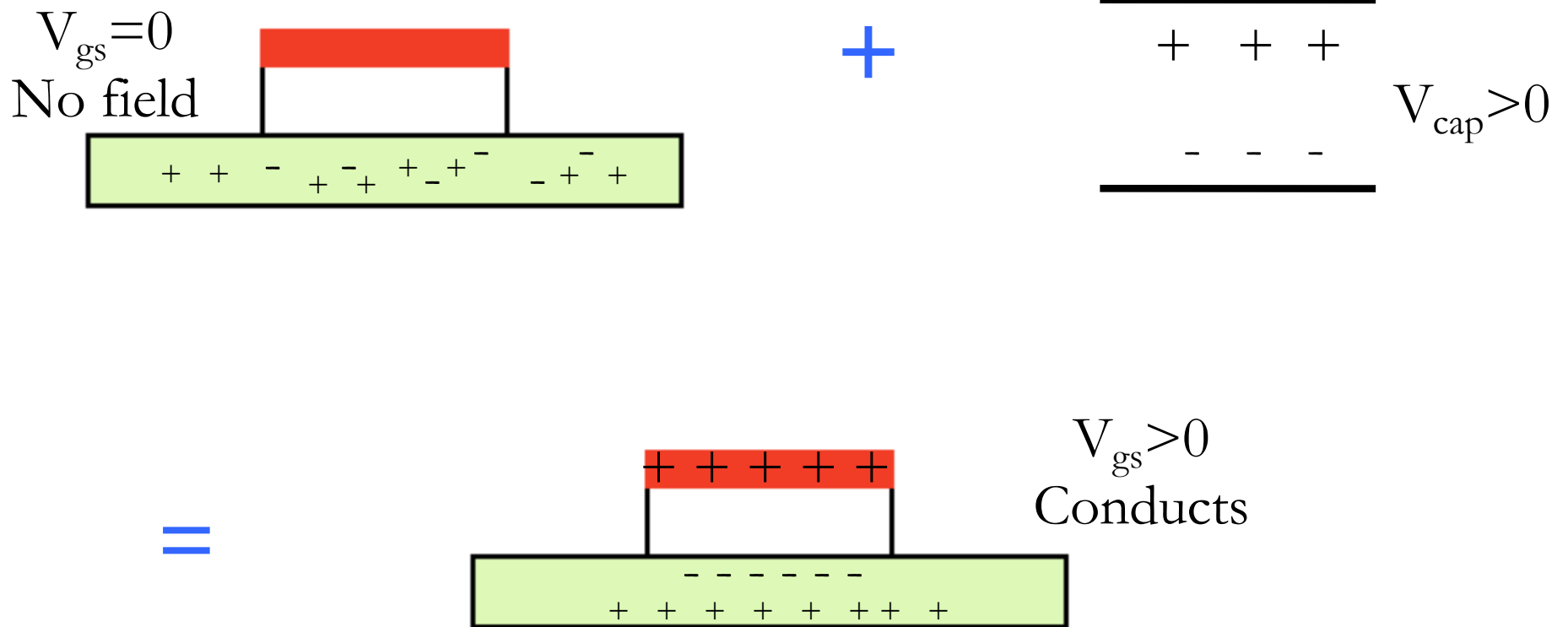
+



$V_{cap} > 0$

# MOS Field?

- What does “capacitor” field do to the doped semiconductor channel?

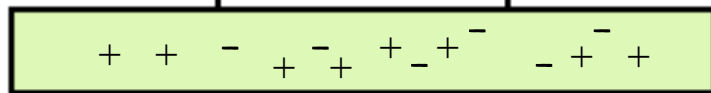




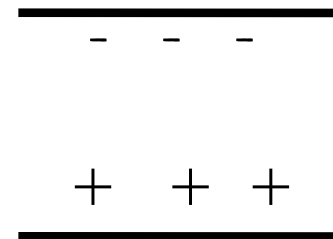
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No field



+

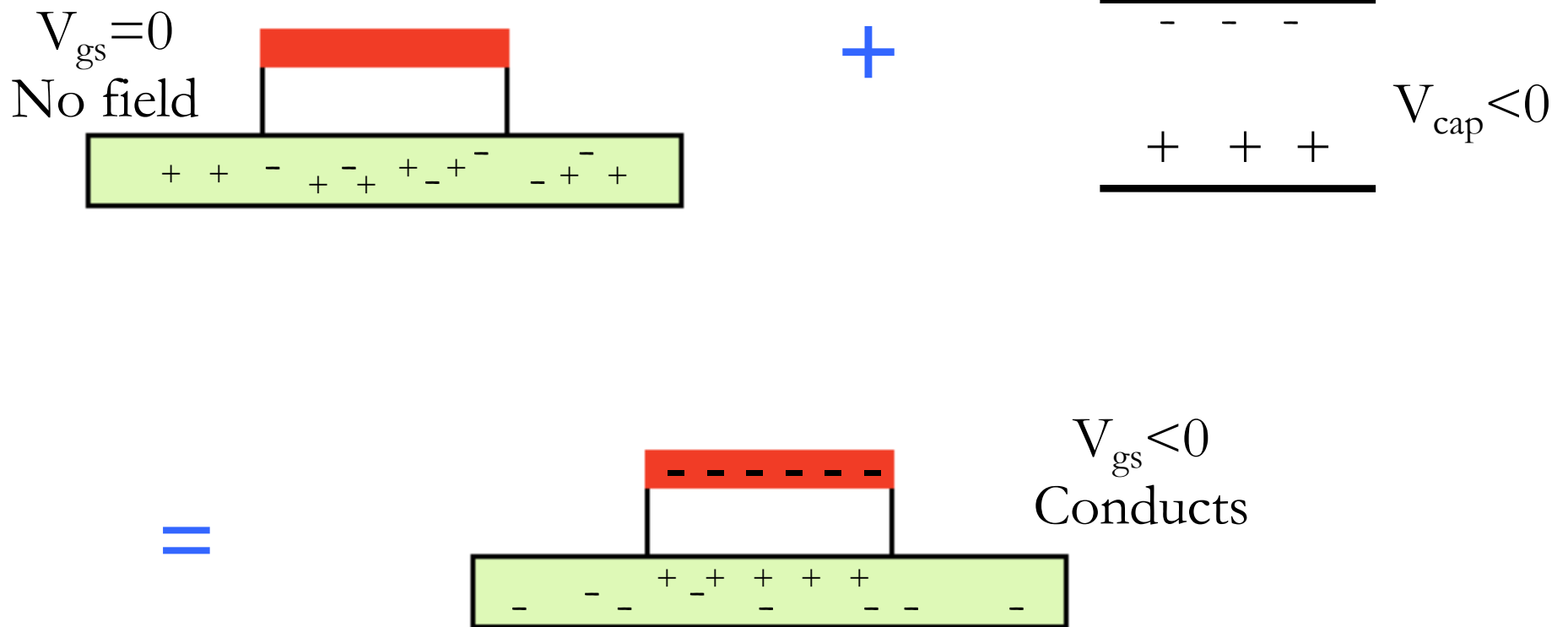


$V_{cap} < 0$



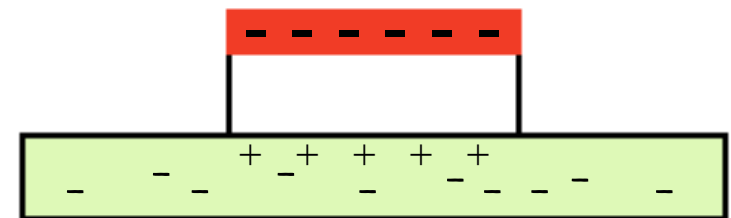
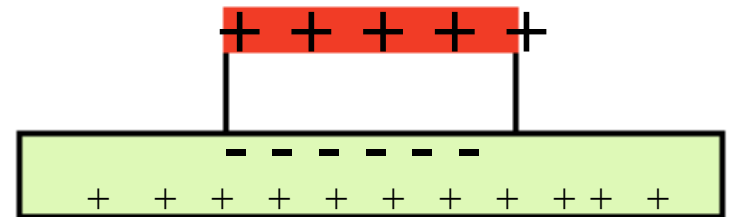
# MOS Field?

- What does “capacitor” field do to the doped semiconductor channel?



# MOS Field Effect

- Charge on capacitor
  - Attract or repel charges to form channel
  - NMOS: Positive field on p-type substrate
    - Attracts electrons to surface to form conducting channel
  - PMOS: Negative field on n-type substrate
    - Attracts holes to surface to form conducting channel



# PN Junction

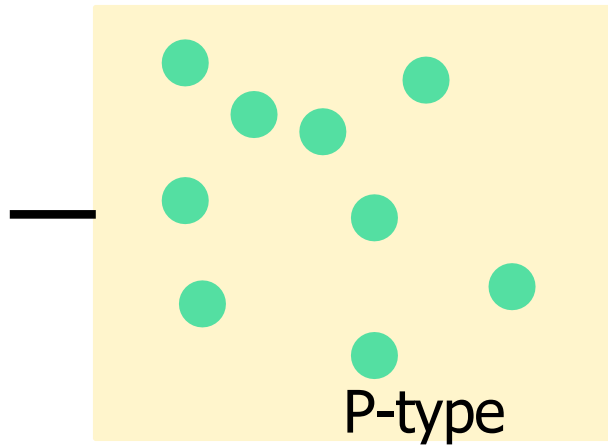
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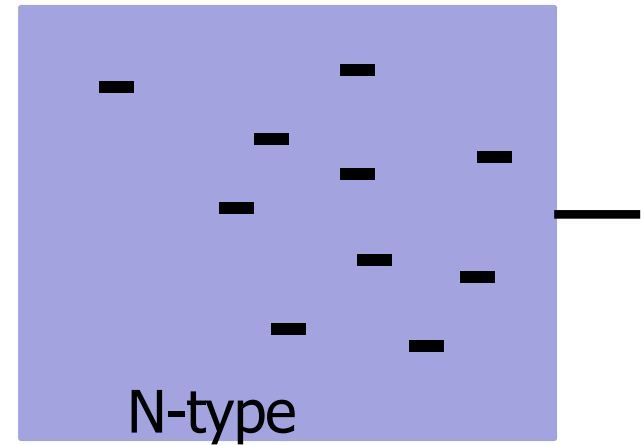


# Doped Silicon

● = hole  
- = electron

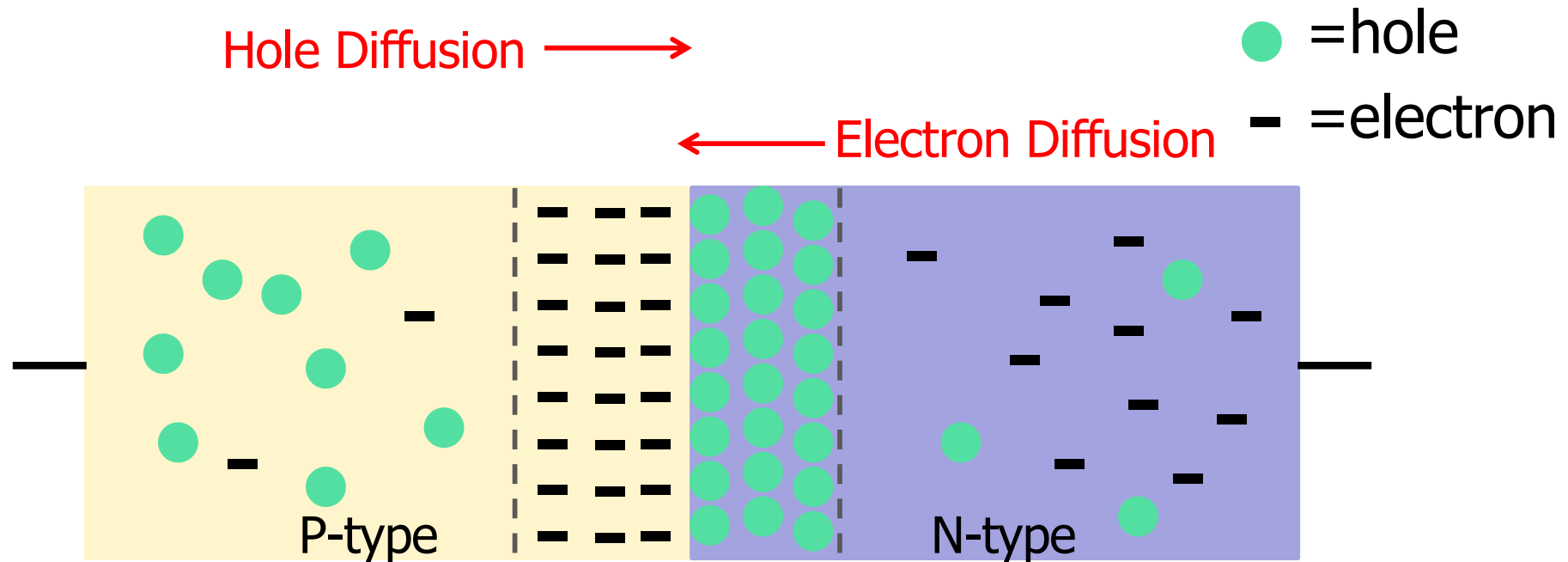


Excess holes



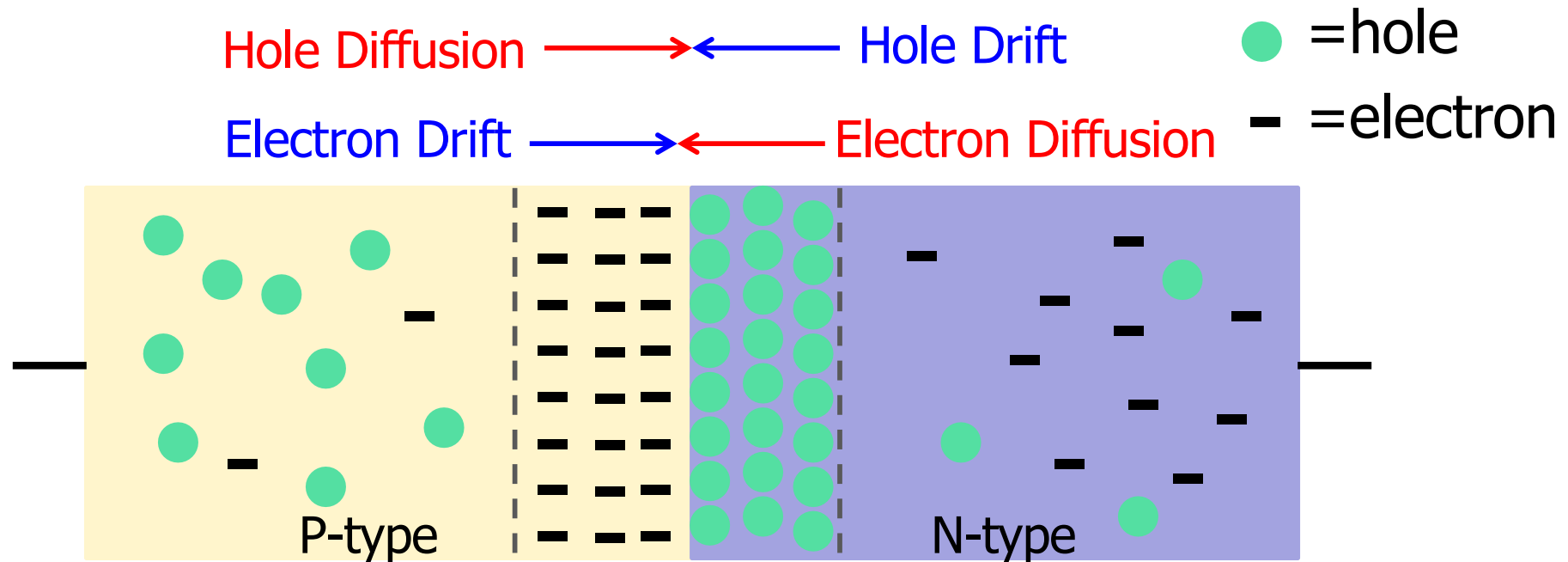
Excess electrons

# PN Junction



- PN junction causes a depletion region to form
  - Electrons diffuse from N-type to P-type
  - Holes diffuse from P-type to N-type
    - Diffusion current caused by diffusion of carriers

# PN Junction



- PN junction causes a depletion region to form
  - Electrons diffuse from N-type to P-type
  - Holes diffuse from P-type to N-type
    - Diffusion current caused by diffusion of carriers
- Equilibrium achieved when  $V_{bi}$ , built-in potential, is formed across the depletion region
  - Drift current caused by E-field due to  $V_{bi}$  to counteract diffusion current

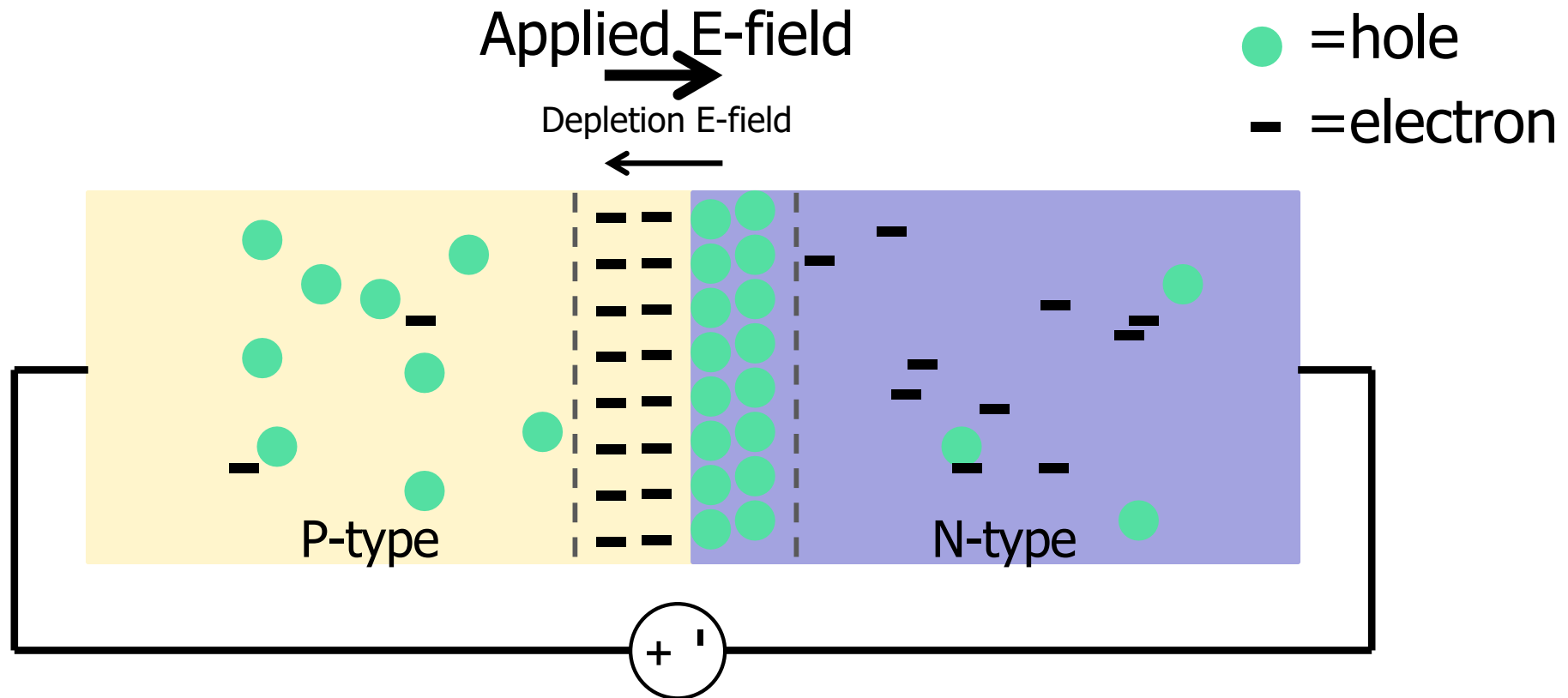


# Drift/Diffusion Currents

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- Diffusion current
  - Current caused by semiconductor diffusion of holes and electrons
  
- Drift current
  - Current due to movement of holes and electrons caused by force from potential difference induced e-field

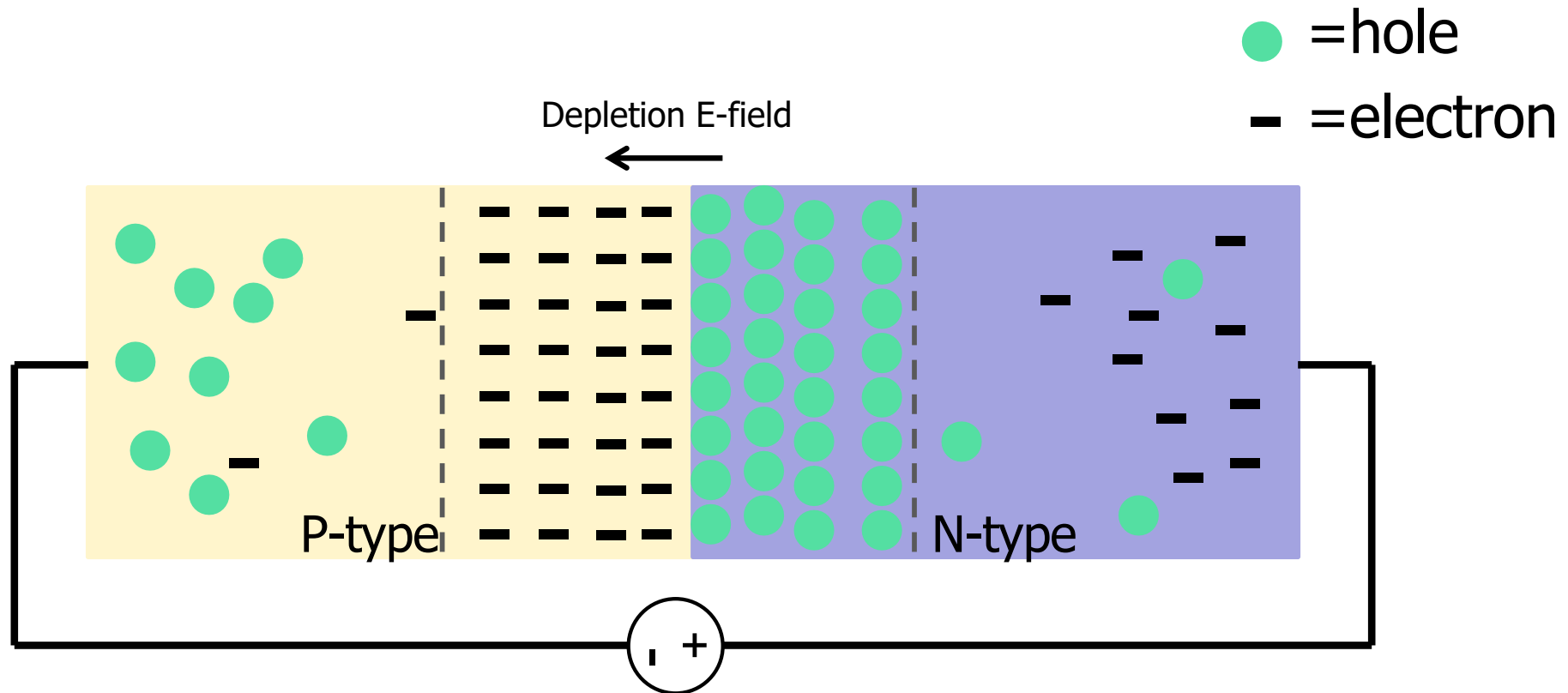
# PN Junction – Forward Biasing



- Forward biasing connect positive terminal to p-type and negative terminal to n-type
  - Holes/electrons pushed towards depletion region, causing it to narrow
  - The applied voltage e-field continues to narrow the depletion region (i.e reduce the depletion e-field)
  - current flows through the device from p-type to n-type



# PN Junction – Reverse Biasing



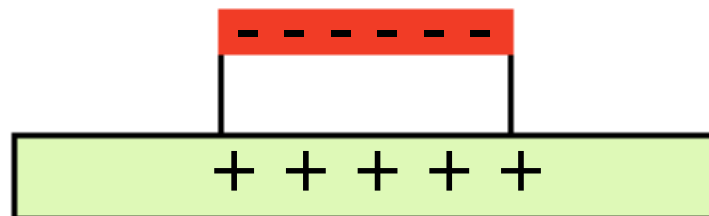
- Reverse biasing connect positive terminal to n-type and negative terminal to p-type
  - Holes/electrons attracted away from depletion region, causing it to widen
    - No current flows through the device
  - If reverse bias increases past breakdown voltage, the depletion e-field increases until breakdown occurs and reverse biased current flows causing thermal damage to junction



# Big Ideas: MOSFET

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- Semiconductor can act like metal or insulator
- Use electric field to modulate conduction state of semiconductor





# Admin

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- HW 2 due Monday
  
- More Fabrication Videos:
  - From sand to silicon (intel) -  
<https://www.youtube.com/watch?v=Q5paWn7bFg4>
  - How microchips are made -  
<https://www.youtube.com/watch?v=F2KcZGwntgg>