



High Rate Growth of SiO₂ by Thermal ALD Using Tris(dimethylamino)silane and Ozone

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Simply ALD

Introduction: Thermal ALD of SiO₂

Many Existing SiO₂ ALD Processes are Unsatisfactory

A few examples

(1) TEOS Process, tetraethoxysilane, Si(OCH₂CH₃)₄

- Reacts with H₂O when catalyzed by NH₃ or amines, even at room temperature
- 0.07-0.08 nm/cycle at 300 K

J. D. Ferguson,^a E. R. Smith,^a A. W. Weimer,^b and S. M. George, *J. Electrochem. S.*, **151** (8), G528-G535 (2004)

- Too slow, required large exposures ($\sim 10^{10}$ L) for surface reactions to reach completion (1 L = 10^{-6} Torr-s)

Introduction: Thermal ALD of SiO₂

(2) 3-aminopropyltriethoxysilane process, O₃ and H₂O:



- Self-catalyzed hydrolysis due to existence of amino group
- 120-250°C, GPC=0.05~0.06nm/cycle

J. Bachmann, R. Zierold, Y. T. Chong, R. Hauert, C. Sturm, R. Schmidt-Grund, B. Rheinlnder, M. Grundmann, U. Gosele, and K. Nielsch, *Angew. Chem. Int. Ed.* **47**, 6177-6179 (2008).

- High quality SiO₂
- Slow process (requires two oxidants H₂O and O₃)
- Unreacted precursor caused frequent damages to vacuum pumps

Introduction: Thermal ALD of SiO₂

(3) TDMAS (or 3DMAS)-H₂O₂ Process:

tris(dimethylamino)silane, [(CH₃)₂N]₃SiH

- First reported to be a reactive precursor with H₂O₂ oxidant

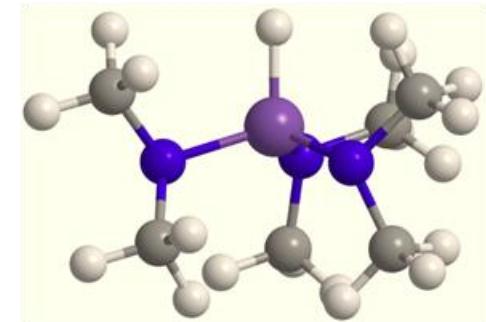
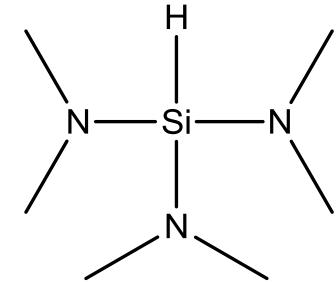
B. B. Burton, S. W. Kang, S. W. Rhee, and S. M. George, *J. Phys. Chem. C* 113, 8249–8257 (2009)

- Precursor can be used in a wide temperature range: 150-550°C
- Inability of H₂O₂ to remove all Si-H bonds at lower temperatures
usable temperature $\geq 450^\circ\text{C}$
- Low thermal stability of H₂O₂

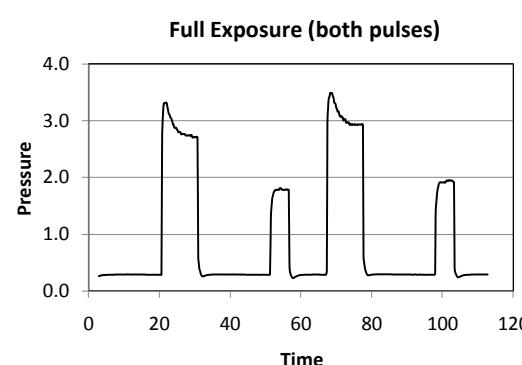
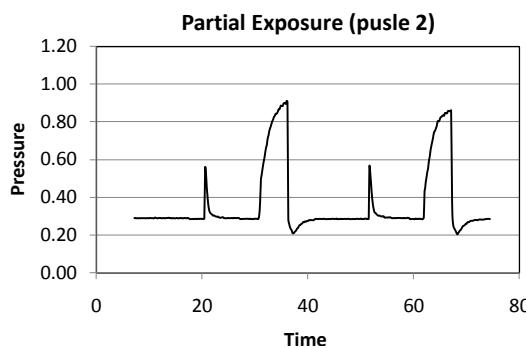
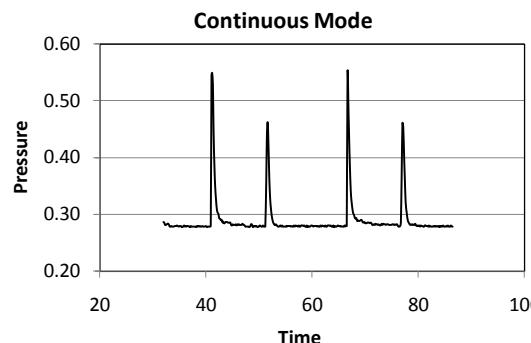
The TDMAS – O₃ Process

TDMAS, [(CH₃)₂N]₃SiH

- High vapor pressure at ambient temperature
BP=145-148°C, 16mmHg at 4°C
no heating needed
- Insoluble in H₂O
- No reaction with H₂O or O₂ up to 350°C



Three Different Reaction Modes



Cambridge NanoTech's ALD Reactors

(1) Continuous Mode

- Normal pulse/purge steps for all precursors

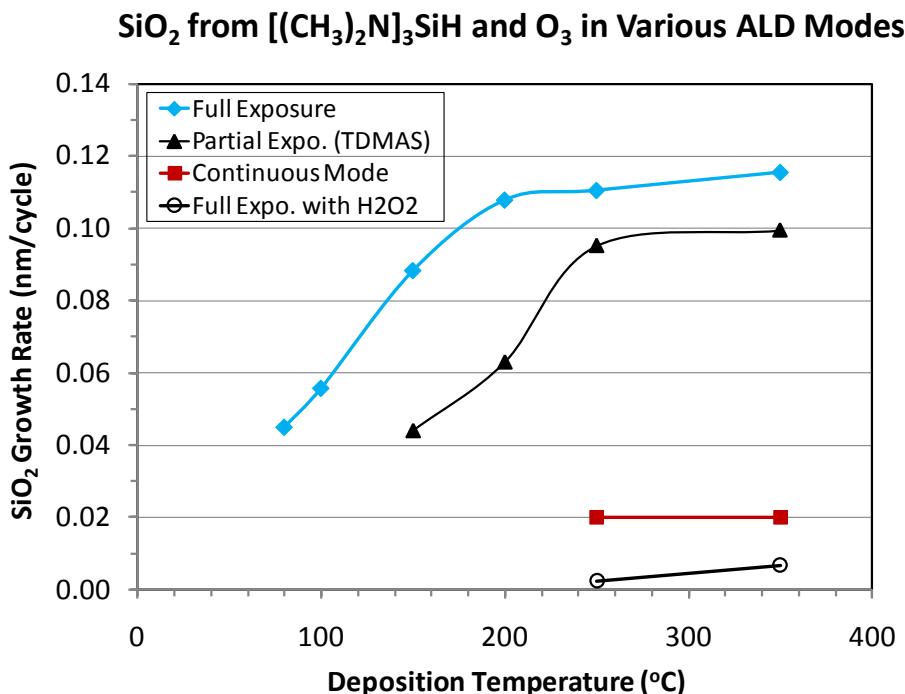
(2) Partial Exposure Mode

- One precursor with an extra hold step (pulse/hold/purge), staying in reactor longer to increase residence time
- Different from an extended pulse time

(3) Full Exposure Mode

- All precursors with pulse/hold/purge steps

The TDMAS – O₃ Process



Continuous:

very low growth rate
~ 0.02 nm/cycle

Partial Expo:

TDMAS exposure (28 sec.) key to high growth rate

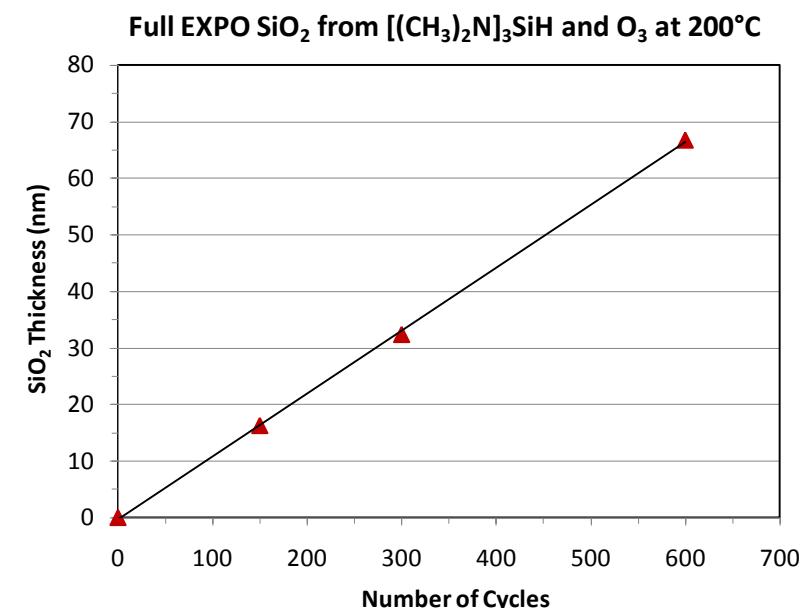
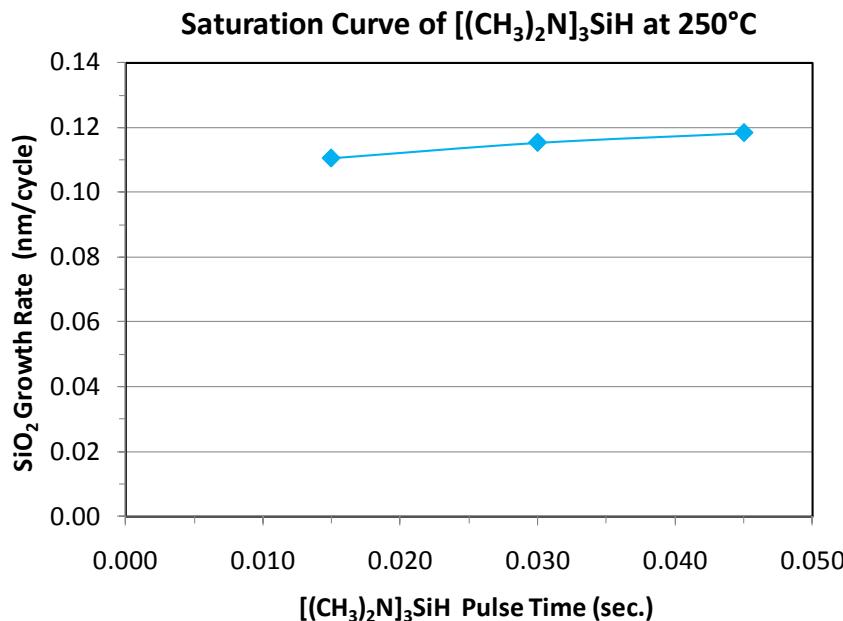
Full Expo:

O₃ exposure (7 sec.) further increased growth rate

Growth rate much lower with H₂O₂

TDMAS Saturation and Linear Growth

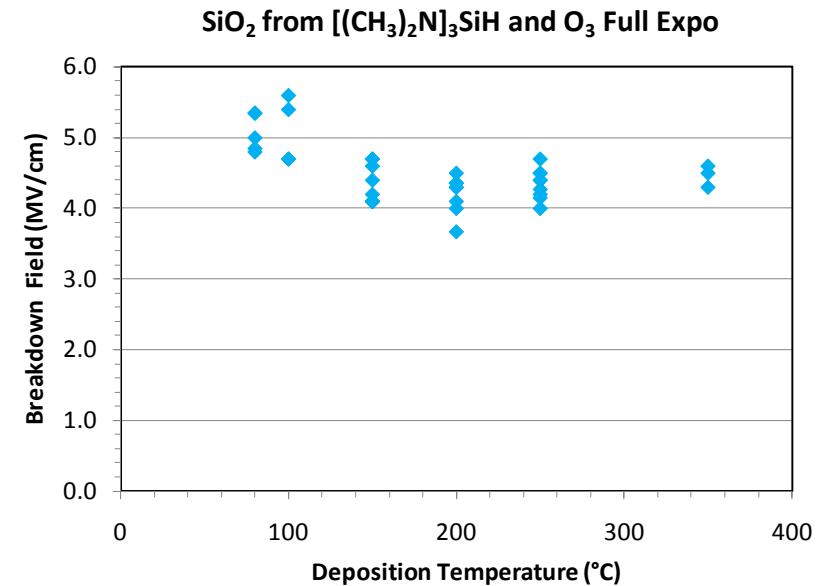
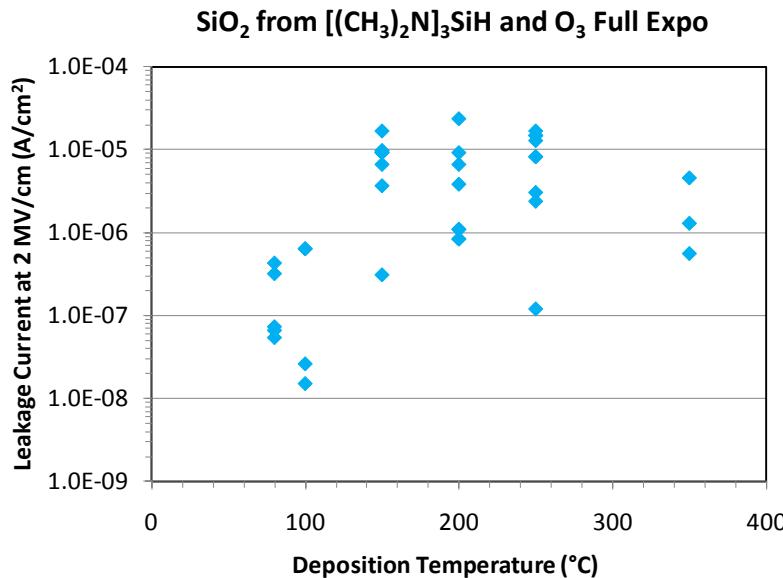
- Deviation from a flat saturation curve with TDMAS dose could be due to less than full saturation from limited exposure time
- Growth is linear without nucleation delay
- Process completely repeatable



Electrical Properties by mercury probe

Leakage Current and Breakdown Strength

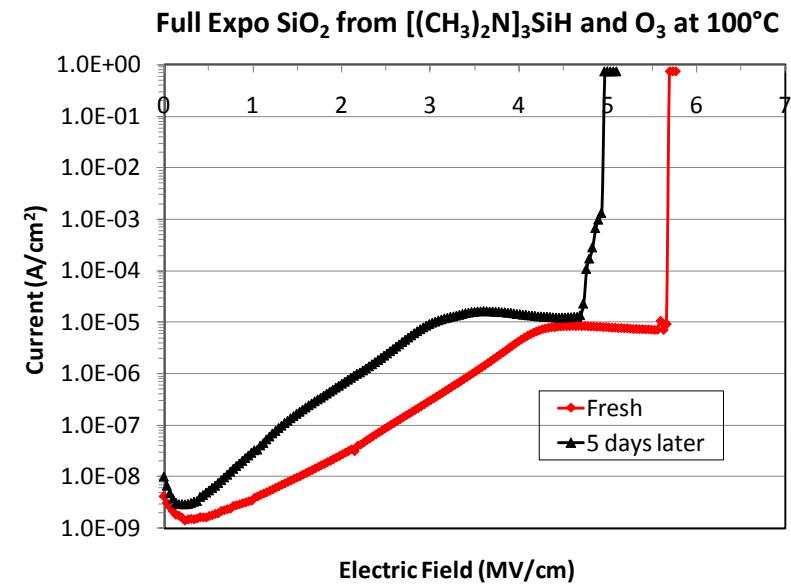
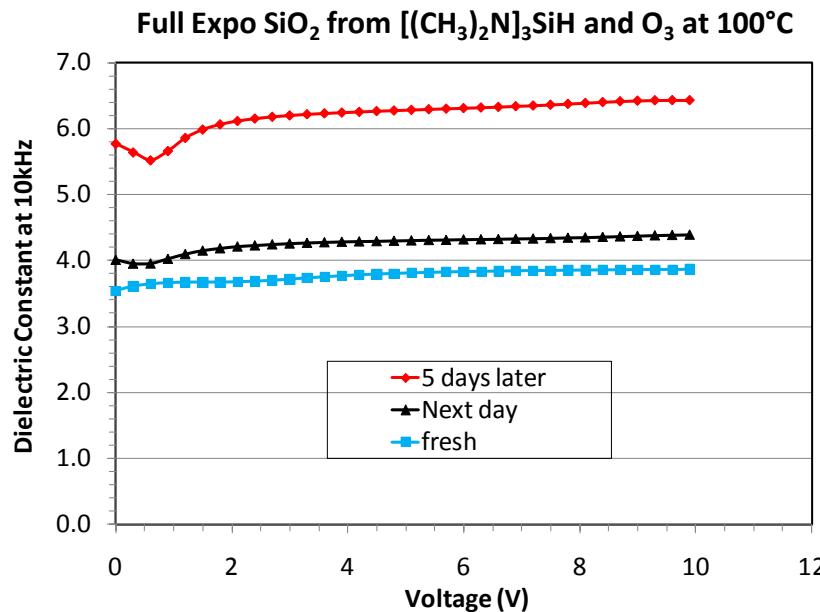
- Data scattered due to changing properties with time
- 80-120°C process produced SiO₂ films with lowest leakage current and highest breakdown strength; 200°C the worst



SiO_2 films absorb H_2O in air

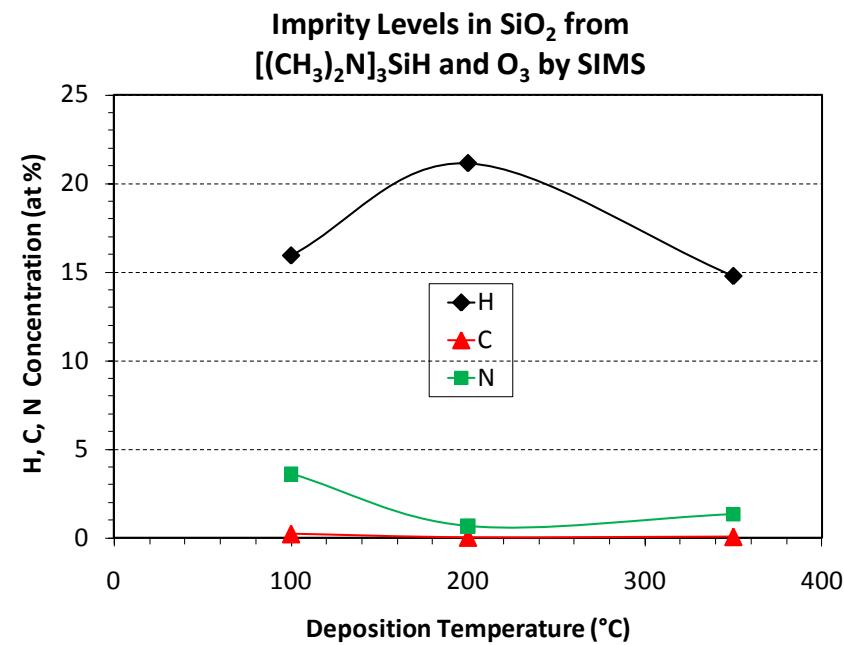
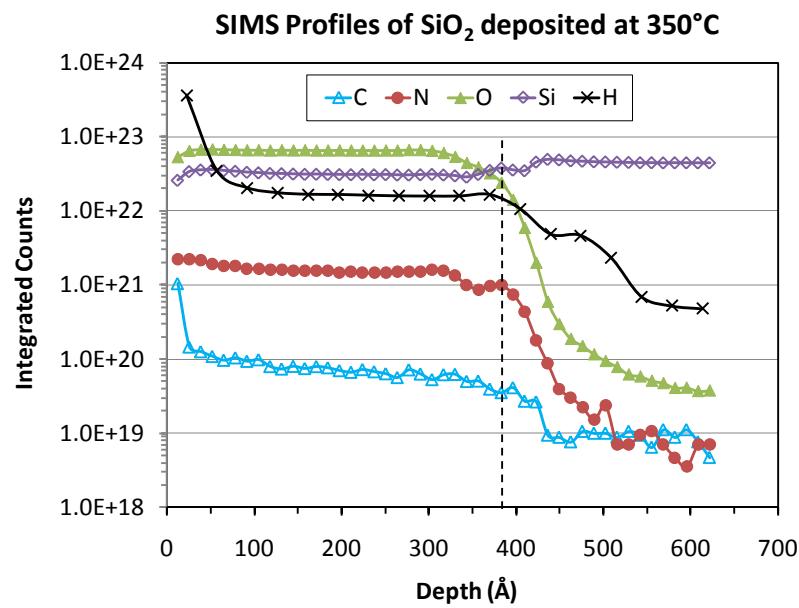
Change in Dielectric Constant, Leakage Current and Breakdown Field with Time

- Dielectric constant of fresh 100°C SiO_2 close to ideal value of 3.9
- It increased with air exposure time until reached a saturated state



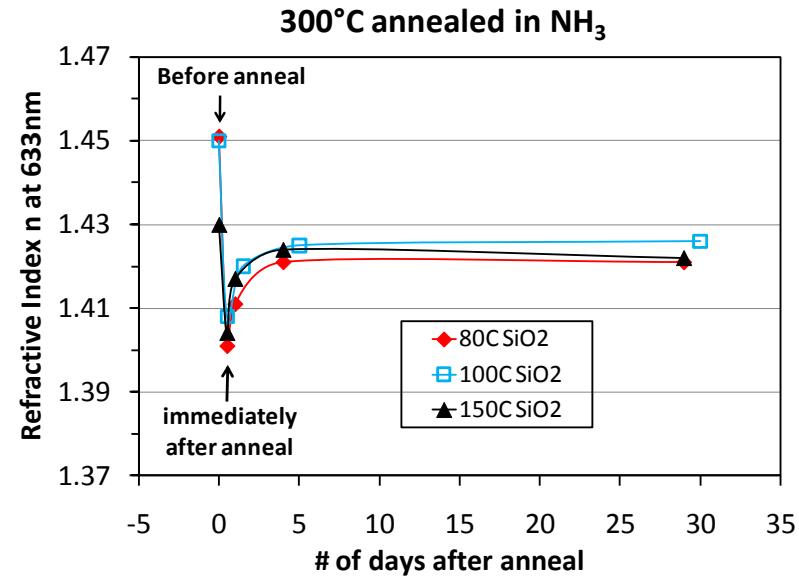
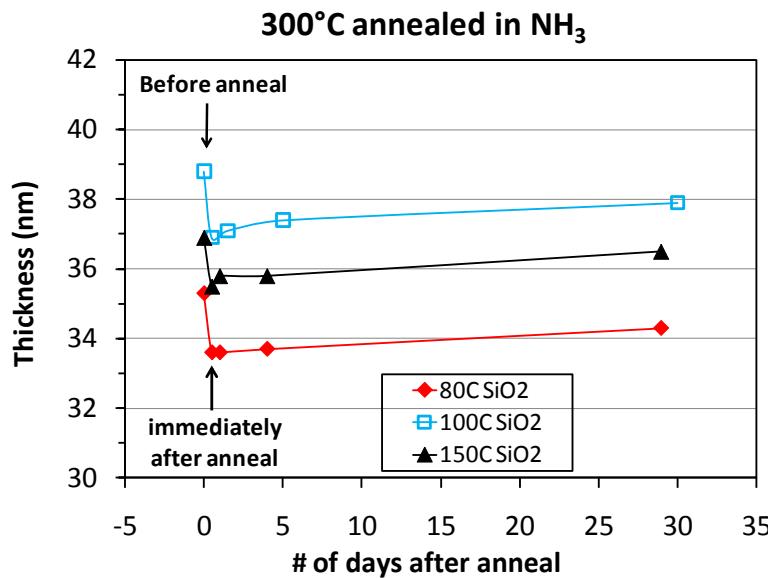
SIMS Analysis

- Very Low C impurity (highest at 0.2 at% with 100°C SiO₂)
- Low N at high temperatures (highest at 3.6 at% with 100°C SiO₂)
- High concentrations of H (highest with 200°C SiO₂)



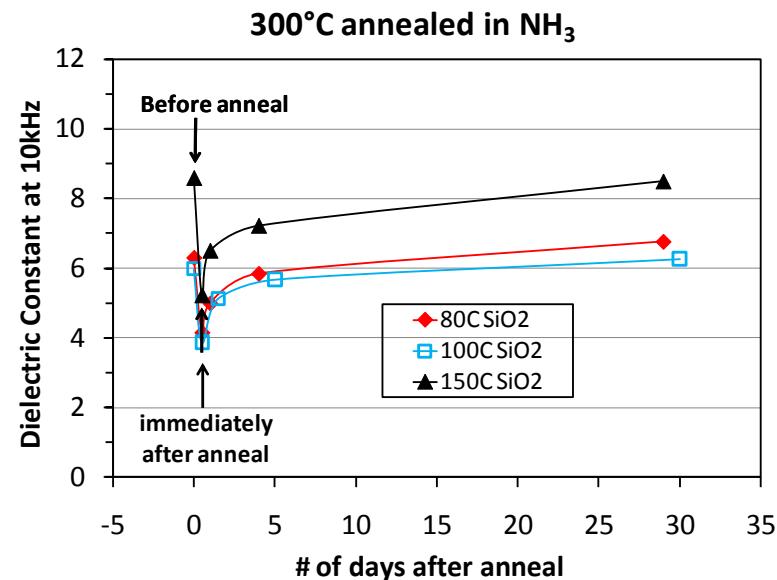
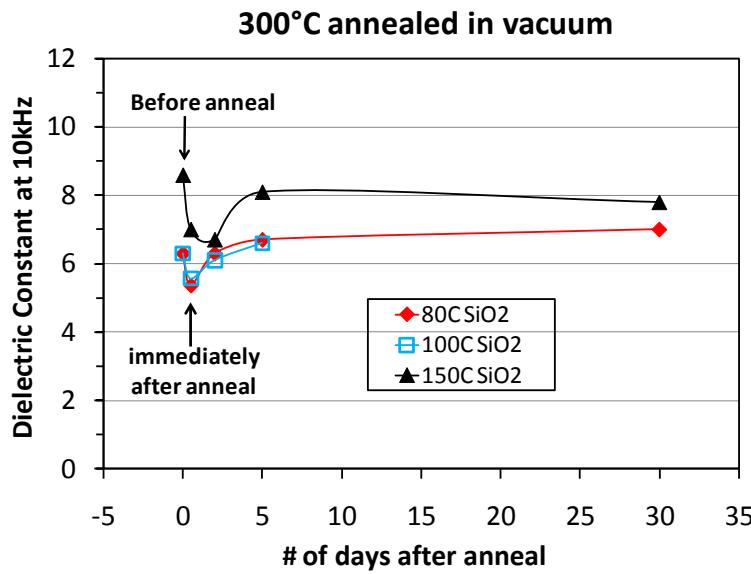
H_2O -saturated SiO_2 films: annealing in vacuum or NH_3

- Annealing at $\geq 300^\circ\text{C}$ led to 3-5% decrease in thickness and a slight drop in refractive index
- Refractive index partially reversible upon re-exposure to air



H_2O -saturated SiO_2 films: annealing in vacuum or NH_3

- 300°C anneal in NH_3 was more effective than vacuum anneal
Dielectric constant dropped to 3.9 for NH_3 -annealed 80-120°C SiO_2
- NH_3 facilitates removal of OH groups?
- Reversible dielectric constant with re-exposure to air



Summary

- **TDMAS-O₃ a good ALD process for SiO₂**
- **Full exposure mode helps growth saturation**
- **80-120°C SiO₂ films have better electrical properties**
- **Uncapped ALD SiO₂ absorbs H₂O in air**
- **Annealing of H₂O-absorbed SiO₂ films in NH₃ at 300-350°C restores electrical properties**